BERLINER GEOWISSENSCHAFTLICHE ABHANDLUNGEN



Reihe A Band 176-II

Jörg Hettler and Bernd Lehmann

Environmental Impact of Large-Scale Mining in Papua New Guinea: Mining Residue Disposal by the Ok Tedi Copper-Gold Mine



FU • TU • TFH Berlin 1995

BERLINER GEOWISSENSCHAFTLICHE ABHANDLUNGEN

Reihe A: Geologie und Paläontologie · Reihe B: Geophysik · Reihe C: Kartographie Reihe D: Geoinformatik · Reihe E: Paläobiologie

Reihe A: Geologie und Paläontologie

Herausgegeben von geowissenschaftlichen Instituten der Freien und der Technischen Universität Berlin sowie von der Technischen Fachhochschule Berlin

Schriftleitung: Dr. Ch. Kuhnert (FU), Prof. Dr. U. Ripke (TFH). Dr. E. Schrank (TU), Prof. Dr. H. Keupp (FU)

Für den Inhalt der Beiträge sind die Autoren allein verantwortlich.

ISBN 3-89582-019-9 · ISSN 0172-8784

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Printing of this publication was funded by United Nations Environment Programme (UNEP)

> Druck: G. Weinert, Offsetdruckerei Saalburgstr. 3 12099 Berlin Germany

Environmental Impact of Large-Scale Mining in Papua New Guinea: Mining Residue Disposal by the Ok Tedi Copper-Gold Mine

by

Jörg Hettler and Bernd Lehmann **

Summary

The Ok Tedi copper-gold mine, located in the castern part of the central mountain range of New Guinea, discharges daily approximately 80,000 tons of ore processing residues and a similar volume of waste rock and overburden into the headwaters of the Ok Tedi River. The Mount Fubilan orebody, which is the source of heavy metal-rich sediments deposited along the Ok Tedi and on the Fly River floodplain, contains a suite of base metals, of which copper is of primary environmental concern. The Ok Tedi flows into the Fly River 200 km downstream of the discharge point, where the mining wastes carried as suspended load are diluted. In the present study, the deposition of mine-derived sediments in the lower part of the Middle Fly River floodplain, and the hydrochemistry and potential mobilization of trace metals, particularly copper, in the alluvial plain was investigated. To this end, a total of 156 sediment cores and surface sediment samples and 117 water samples from the upper Ok Tedi and the Middle and Lower Fly River floodplain were taken. The suspended matter content in Middle Fly River water today is about 5-10 times higher than the natural background of about 60 mg/l. Near-surface sediments deposited along the river channel contain up to 1100 mg/kg Cu, the mean value is $530 (\pm \sigma = 240-820)$ mg/kg.

Of the floodplain water bodies, cut-off meanders receive the largest quantities of mine-derived sediment. Deposits of up to 70 cm in thickness of copper-rich material with 800-1000 mg/kg Cu, which have accumulated since the mine started discharging residues in 1984, were detected in oxbow lakes. Very high deposition rates (around 4 cm/year) of mine-derived sediment were determined in locations close to the creeks and channels which link the Fly River with the outer floodplain. Due to the flat terrain, turbid Fly River water intrudes regularly upstream of the floodplain tributaries (measured intrusions up to 25 km). A thin layer of 1-5 cm of copper-rich material (400-900 mg/kg Cu) was usually found on the bottom of drowned (tributary) valley lakes. Copper in sediment deposited in the pre-mining period gave a median value of 44 ($\pm \sigma = 25-63$) mg/kg. Riverine particulate matter also settles down on the floodplain at times of overbank flow, which leads to extensive copper contamination in low-lying swamp sites close to the river. Natural deposition rates in the floodplain were determined by C^{14} age dating to range between 0.1-1 mm/year, with the exception of oxbow lakes, where natural sedimentation is much higher. Leaching of copper from material deposited on swampy, vegetated floodplain sites was detected in sediments which were also strongly depleted in calcite and sulfide, which are the components most easily mobilized of mine-derived material. The variable water table, oxidizing environment and acidic conditions generated by decomposing vegetation, typical features of low-lying floodplain swamps, facilitate the mobilization of copper from the solid into the dissolved phase. Water taken from swampy floodplain locations showed copper values of up to 50 μ g/l Cu in the filtered sample (membrane filter 0.45 μ m).

^{*} Final report on a study funded by the United Nations Environment Programme (UNEP)/SPREP

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Average copper content in mixed waters of the inner floodplain is around 9 ($\pm \sigma = 5-14$) µg/l; the Fly River water has around 17 ($\pm \sigma = 13-19$) µg/l Cu. Copper concentrations in unpolluted floodplain waters were measured at below 2 µg/l. Nearly all dissolved copper in the Fly River system is complexed by dissolved organic carbon compounds, however, a fraction of these complexes appears to be labile and reactive. Comparison of dissolved copper levels measured during the present study in the Middle Fly River floodplain with literature data on copper toxicity and international water quality guidelines shows that chronic toxicity of the metal to the aquatic community is to be expected. Significant negative ecological effects, particularly on the local fish population, may develop with a considerable time lag, since aquatic organisms at the base of the food web are the biota most sensitive to copper.

Foreword

by Dr. David Mowbray, Professor of Environmental Science at the University of Papua New Guinea and Government Adviser to the PNG Department of Environment and Conservation

The Ok Tedi mine in PNG ranks amongst the world's largest copper mines. It is a major contributor to the PNG economy accounting for up to 45% of PNG's total annual export revenues. The PNG Government is a 30% shareholder in Ok Tedi Mining Ltd. (OTML). Mining for gold commenced in 1984, and for copper in 1986. Mining for copper is expected to continue for a further 10-15 years.

Initially it was intended that a tailings dam should be constructed to contain the mine's waste. However for economic reasons and given that the area is structurally unstable, no tailings dam was constructed, and so since 1986 from 100,000 to 150,000 tonnes of waste per day have been discharged into the Ok Tedi and Fly River system. In 1989 the PNG Government set compliance standards covering suspended particulate load, dissolved and particulate copper levels, biological parameters like fish catch, and others. It is argued by some that the environmental management of the mine is based on monitoring for compliance with standards that have been established "to facilitate the realization of OTML's mining schedule" rather than to protect the environment.

Ok Tedi Mining Ltd. has an extensive environmental monitoring programme. Assessment of compliance monitoring and interpretation are done routinely by the company (and the Government), but no detailed data are available for public scrutiny (if at all) for at least a year after data are collected. This lack of public accountability for environmental performance is unacceptable.

Inevitably the Ok Tedi mine evokes different emotions and perceptions by different groups in PNG. For example, Mr Kipling Uiari, the former Deputy General Manager of OTML and now BHP PNG General Manager, stated in 1993 at the 20th Waigani Seminar on "Environment and Development" here at the University of PNG:

"Importantly, the mine has brought positive changes to the quality of life of neighbouring communities in the previously underdeveloped Western Province. Investment in project infrastructure has amounted to about K300m and many of the basic services now established are available to all citizens in the area. This is particularly true of the road, power, health and water systems, transport and communication facilities. These services are meeting the basic needs of local villagers by increasing life expectancy and providing access to education, employment and business opportunities.

Ok Tedi Mining Ltd. has taken a number of iniatives to encourage as much participation as possible from people in the region and so ensure human development goes hand in hand with mineral development.

Sustainable development also requires good environmental stewardship. During the devlopment of the Ok Tedi project there have been criticisms of its environmental impacts, many of them based on inaccurate information. Ok Tedi Mining Ltd. is a responsible resources development company which recognises that all environmental effects must be carefully considered. It is the extent of these impacts and whether they are reversible or not,

which must be considered in relation to the social and economic benefits to the communities involved and the nation as a whole."

Whereas at the same seminar Mr Alex Maun, a prominent landowner stated:

"With the mining operation our life has changed. The loss of our rainforest and degradation of the environment cannot be calculated in terms of short-term money. We are rural village subsistence farmers who depend on the environment for survival.

Before the Ok Tedi Mine operation our life was paradise, we enjoyed both the aquatic and terrestrial resources. We used the river for fishing, washing, drinking and transportation. We made gardens near the river banks which lasted 3 to 5 years.

Now we river people can no longer drink from the river nor can swim, bath or wash clothes in the river. We lack the protein in our diet that was formerly provided by the aquatic and terrestrial resources. Overflow of the Ok Tedi River has caused the wild life near the river banks and floodplains to disappear. Some game animals were drowned by sudden floods. Trees in the floodplains are dying completely forcing the wild life to migrate to other areas. Now gardens are no longer made near the river banks.

I would like to stress that Ok Tedi Mining is causing irreversible destruction along the Fly River an Ok Tedi River. Whatever it is destroying will never come back to normal, e.g. customary land, sago swamps, etc. OTML is bringing unsustainable development. We affected river people think it is nonsense talking about sustainable development when the mess done by the Ok Tedi Mining is not cleaned up."

In 1994 Ok Tedi again hit the headlines: "Ok Tedi has to build tailings dam - Zeipi", so says *Post Courier* headline on 18 March 1994. The report goes on:

"Ok Tedi Mining Ltd (OTML) will be told to build a series of dams to dispose of mine wastes or "ship out" if it refuses. Mr Zeipi has always insisted a tailings dam be built because, he says, the river dumping has damaging effects on the ecosystems."

(Mr. Zeipi is Minister for Environment and Conservation).

"K2b case looms on mine damage" again says *Post Courier* headline, this time on 4 May 1994. Then in *Post Courier* on 6 May a report states:

"The leader of a Papua New Guinean clan lodged a writ in the Melbourne Supreme Court yesterday against Australia's biggest company, BHP, seeking unspecified damages for allegedly poisoning the Ok Tedi River and destroying his people's subsistence way of life. It also alleges the PNG Goverment, a 30 percent shareholder in Ok Tedi Mining Ltd., had "failed, neglected and refused" to enforce environmental agreements and convenants."

These matters are still being dealt with in both PNG and Australian courts and have not been resolved. Furthermore present debate continues between Government ministers and leaders and Mr Zeipi on the type of compensation payments for environmental damage and whether a tailings dam should and could be built.

The Floodplains

The Ok Tedi mine adds about 58 million tonnes of sediment to the Ok Tedi River per year. It is estimated that 30% is deposited along the Ok Tedi, much of the rest reaches the Fly Delta. An unknown amount is deposited along the Fly River and in the floodplains in the middle Fly. Studies on the coastal and marine ecosystems in the Fly Delta, Papuan Gulf an Torres Straits are being done by both Ok Tedi and Australian scientists funded by Ok Tedi. Studies on the floodplains of the middle Fly are being done by Ok Tedi scientists, and are also the focus of the present UNEP study.

OTML, in its booklet "Ok Tedi, the Environment and You" states that

"copper in the sediment in the Fly River is also being transported during floods onto the floodplain where it settles into the lakes, streams and the flooded forest. At this time there is very little scientific information to tell us what effect this copper on the floodplain might have on fish life, feed and breed in this area".

Consequently OTML has commissioned studies of the amounts in and effects of copper on the floodplain ecosystems. Unfortunately the PNG Government and independent researchers generally lack the financial and human resources to do independent monitoring and research of depth and detail. Hence most studies are done by or commissioned by OTML.

This study by Jorg Hettler and Bernd Lehmann is valuable since it is done by overseas scientists funded by UNEP through the universities of Berlin and Clausthal and the University of PNG. It is apparent from their work that copper is being deposited in the floodplain at higher than expected levels. Hopefully future independent scientists can assist in deciding if this copper contamination is likely to have biological effects, for examples if fish populations may be affected.

The present study is a valuable contribution to our present state of knowledge of the Ok Tedi/Fly River system. In fact more environmental research has been done on this tropical river system over the last 15 years than probably any other tropical river system in the world. Yet there is still much uncertainty!

Back in 1982 in his book "The Pot of Gold", Richard Jackson made a statement which (in slightly modified form) still is applicable 13 years later:

"In our present state of knowledge of the workings of natural systems, it is only in the rarest of occasions that honest environmental experts agree as a body and confidently predict a sequence of future events and outcomes. The Ok Tedi project is not one of those occasions. Decisions have been made in the project hoping that environmental risks will be worth it, that is, hoping that in the light of the facts available the magnitude of the environmental impact will be minimal. But the facts available are, still, too few. In the Ok Tedi case, the experts, if they are honest, will be keeping their fingers crossed and environmental monitoring systems under constant scrutiny ..."

and hoping that the environmental damages remain acceptable!

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1. INTRODUCTION

The environmental problems of mining activities have been receiving increasing attention worldwide in the last decade. The shift in the search for raw materials outside of the classical producer countries to the world's marginal regions like tropical rainforests has led to a number of environmental and social problems. In many cases, the extraction of natural resources has been the first industrial activity being developed in peripheral regions. Requirements of infrastructure, logistics and workforce may completely change the environmental and social patterns in regions which have previously been nearly untouched by man. One of the best documented examples of such a development is the Ok Tedi gold-copper mining project in Papua New Guinea. Before the economic value of the Mount Fubilan orebody was discovered in the late 1960's, the mountain was a sacred place to the sparse population of the local Wopkaimin Papua tribe, who were inhabiting one of the world's most isolated regions. The environmental impact of the mining operation, located in a difficult physical environment, has been subject of controversy since the project's early planning stages. The fact that the Ok Tedi mine works without waste retention facilities has been in the focus of the debate in the last few years. In May 1994, local inhabitants living along the Ok Tedi River lodged a \$A4,0000 million compensation claim against the mine's operator, BHP, over environmental damage caused by the discharge of tailings and waste rock into the Ok Tedi. (Mining Journal , May 6, 1994)

The mining company is engaged in an extensive environmental monitoring program and also has commissioned a number of studies on individual problems which have been executed by international research institutions and consultants. The present study, funded by the United Nations Environment Programme, is the first independent research project undertaken in the area. It investigates one of the most important environmental aspects of the mining project. Based on discussions with the Environment Department of Ok Tedi Mining Ltd., it focusses on the fate of mine-derived sediments deposited in the Fly River floodplain.

The project's coordinator was Dr. Bernd Lehmann, Professor of Applied Geology at Technische Universität Clausthal. The responsible research officer was Jörg Hettler, M. Sc. (Geology), of the Department of Environmental and Resource Geology at Freie Universität Berlin. The sedimentological part of the study was coordinated by Dr. Georg Irion, Professor at Forschungsinstitut Senckenberg, Wilhelmshaven. The University of Papua New Guinea offered comprehensive logistical support during the field and laboratory work.

A draft version of this report has been presented by the study team in a series of discussions held in September 1994 in Port Moresby and Tabubil, Papua New Guinea, with Ok Tedi Mining Limited, the Departments of Mining and Petroleum (DMP) and Environment and Conservation (DEC) of the Papua New Guinea Government, and the University of Papua New Guinea.

2. SAMPLING AND ANALYTICAL METHODOLOGY

2.1. Sampling

Sampling locations were determined with a handheld Global Positioning System (GPS) receiver and available topographic maps of scales 1:100,000 and 1:250,000. Water depth and bottom relief were measured with a portable electronic depth sounder with LCD screen.

2.1.1. Sediments

32 surface sediments from the river bank of the Ok Tedi at Tabubil and from a few locations in the Middle Fly region were sampled with a plastic spoon and collected in geochemical sampling paper bags. The large majority of sediment samples (124) taken along the Fly River was recovered with a gravity corer.

Because of the difficult access to swampy, vegetated floodplain sites, the large majority of sediment cores (and water samples) were taken from channels, lakes and small water bodies which were accessible by boat or dugout canoe. The

gravity corer consists of a massive aluminium tube with a backslash (floating) valve mounted on the upper end. Fins are welded to the aluminium body to stabilize the corer during its free fail through the water column. Coring tubes of transparent acrylic glass ("Plexiglas") of 1 m or 1.5 m length were fixed inside of the aluminium shaft. The corer was brought in an upright position above the water and then released for free fail. To increase penetration depth in the bottom sediment, a ring-shaped lead weight of 10 kg was fitted to the aluminium shaft. The corer is hoisted from the bottom of the sampled water body with a rope fixed to the backslash valve, which remains closed. The sediment cores recovered have a diameter of 36 mm and a length of between 20 and 70 cm depending on the nature of the bottom sediment. The cores, which showed no or minimal disturbance, where pushed out of the plexiglas tubes with a rubber stopper and were packed in clean polyethylene bags. On floodplain sites, the sampling tubes were driven by hand into the sediment. After closing the open end with a rubber stopper, the tube was pulled out. The sediment cores recovered were usually in the same range of lengths as the cores from water bodies.

2.1.2. Water

A total of 117 water samples in the Ok Tedi/Fly River system was taken. Water temperature, pH, conductivity and oxygen content were measured in the field using portable electronic equipment. The reduced species NH_4^+ , HS⁻ and NO_2^- were also determined in the field using "Merck Aquaquant" reagent kits for rapid water analysis, based on a colorimetric method.

All water samples were gulp samples taken by hand 20-50 cm below the water surface, the volumes ranging between 250 and 1500 ml. The wide-mouth polyethylene bottles were soaked in dilute nitric acid for two days and washed with demineralized/distilled water before use in the field. The containers were rinsed twice with water from the sampling site before the sample was taken. Upon return to the field laboratory within a few hours after collection, alkalinity was determined by titration with 0.02 mol HCl to pH 4.5 (APHA Standard Method No. 403).

The water was filtered through 0.45 μ m cellulose nitrate membrane filters (Sartorius, Germany) with a portable "ANTLIA" pressure filtration system (Schleicher & Schuell, Germany) which consists of a pneumatic pump, a 50 ml syringe cylinder and a filter holder of 50 mm diameter. Sometimes a pre-filter (Schleicher & Schuell Blue Ribbon ashless) had to be used for highly turbid water. After sufficient rinsing of the pump and filtration system, a 120 ml water sample was taken and was immediately acidified with 3-4 ml of concentrated nitric acid (Merck Suprapur) per litre of sample. Filters were retained for gravimetric determination of suspended solids. The suspended matter content in some samples was measured using a 1,51 "Imhoff" funnel-shaped sedimentation cylinder. Few water samples were split and a subsample was acidified unfiltered to determine the trace metal content associated with the particulate matter. Water samples selected for anion analysis were filtered but not acidified, DOC samples were stabilized with 2 ml/l of 50% H₂SO₄.

Great care was taken to avoid cross-contamination of collected material. All equipment was acid-washed and rinsed with distilled and deionised water between use for different samples in the field laboratory. Blanks were included and submitted to the same procedure as the samples.

2.2. Analytical Methods

2.2.1. Sediments

Sediment cores were cut in halves in the laboratory with a stainless stell knife to allow visual inspection and taking of photographs.

Sub-sections of sediment cores were wet sieved with distilled water through nylon sieves of 20, 60, 100 and 200 μ m mesh size. The resulting size fractions <20 μ m, 20-60 μ m, 60-100 μ m, 100-200 μ m and >200 μ m were washed from the sieves into plastic containers and dried to constant weight at 80° in a drying oven. Grain size distribution was determined after weighing.

The fraction less than 20 μ m in samples selected for clay mineral analysis was submitted to gravity separation in "Atterberg" sedimentation cylinders. Of the resulting three fractions (<2 μ m, 2-6.3 μ m, 6.3-20 μ m), the finest was used to prepare smear slides which were submitted to X-ray diffraction analysis.

NBS 2704 Buffaio River Sediment	Ka	M	Ņđ	5	N	2 2	u2 e	ខ	4 4	8	Ŋġ	A6	ង	Ŷ	ខ	ĨĨ	-	1	ц ц	- 65 - 65	22	ρ.	80	
Certified Uncertainty	5470 ±140	20000 ±400	12000 ±200	26000 ±300	11600 4	1100 5	52 1 3	æ 0	1.6 161 1.0 ±17		22	23.4 ±0.8	135 ±5		14 ±0,6	# #	8 #	1570 3 1180	60	ar +1 6	14 13 12 13	5 +1 0	38 12 28	
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EDC/BCR RM 280 lake Sediment	Na	F		5	W	2	- 42 - 42	3	8	8	ş	¥9	ង	<u>s</u>	ខ	IN	•	1		1944 1945	a Sr	6.	ßc	
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certified Oncertainty	17600	26000	16430	16870	77500	42360 1	350 29 ±	1 2 4	5 80. 12.80	2 1.6 3 ±0.	1.1	51.0 ±2.4	114 ±4	1.9	20	73.6 <u>1</u> 2.6	102	4040		ία.	81	15	90 12.8 ±0.7	
Analyzed (ICP-S)	15800	24800	16200	16500	77500	1400 1	148 26	2 65	65	1.0	. .	30	87	~	15	19	72	3100]	5	ف 0	81 20	3 14	11 OI	
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Table 1. Quality control with standard reference materials. Values with no uncertainty range given are noncertified. As value with * is by hydride generation/ICP-S.

8

Samples selected for chemical analysis (grain size <20 μ m was the standard fraction used) were shipped to commercial laboratories (ACME and XRAL, Canada), where the sediment was homogenized and pulverized in an agate mill. In the "total digestion" procedure, 250 mg of sample is digested with 10 ml HCl0₄-HNO₃-HCl-HF at 200°C to fuming and diluted to 10 ml with diluted aqua regia. The vigorous digestion procedure leads to potential loss of As, Sb and Cr due to volatilization during HClO₄ fuming. The leach, however, is partial for magnetite, chromite, barite, oxides of Al, Zr and Mn. 35 elements were determined by inductively coupled plasma spectrometry (ICP-S), of which 24 are reported in the present study. Arsenic and antimony were additionally determined by hydride generation with aqua regia digestion and ICP-S analysis. Gold together with some other elements were analyzed in selected samples by neutron activation analysis (NAA) in a commercial laboratory (Bondar-Clegg. Canada). Insoluble carbon and sulfur were determined by Leco furnace, in which CO₂ and SO₂ gases are released during combustion and then detected. A 15 % HCl leach prior to combustion was performed to remove soluble sulfate and carbonates. This procedure ensures that only organic carbon and sulfur bound as metal sulfide is detected.

Two reference sediment samples, NBS 2704 Buffalo River Sediment and EEC/BCR RM 280 Lake Sediment, were submitted to the commercial laboratory for quality control. Results are shown in Table 1. Radiocarbon age dating was performed on five samples of peaty sediment from drowned valley lakes at the C^{14} laboratory of Kiel University, Germany.

2.2.2. Water

Water samples were analyzed at a commercial laboratory (XRAL, Canada), at the laboratory of the Department of Environmental and Resource Geology at Freie Universitat Berlin (FUB), and at the laboratory of the Environment Department of Ok Tedi Mining Limited at Tabubil (OTML).

Commercial analysis for metals was by multi-element ICP-S, of which measurements for 13 elements are reported in this study. The trace metals copper, lead, zinc and cadmium, which were present in some samples in concentrations close to the detection limit of ICP-S, were additionally determined by graphite furnace atomic absorption spectrometry (GFAA) at FUB and OTML laboratories.

Major anions (sulfate, chloride, nitrate) were analyzed by suppressed ion chromatography on non-acidified samples. Dissolved organic carbon (DOC) was determined with a "Technicon" auto analyzer.

Quality control for water analysis was more difficult as compared to sediments since no standard reference material was available. Calcium and bicarbonate values, which were determined by two different methods, gave a correlation of 99% in the data set of all measured values. Values obtained for blanks and measurements of the same sample by different methods and laboratories are given in Table 2. The analyses of blanks (demineralized/distilled laboratory water) showed no or minimal contamination, with the exception of zinc. Reported values for this metal have been corrected by subtraction of zinc content in blank samples.

Statistical analysis of sediment and water values was performed using the U.S. Environmental Protection Agency 'Geostatistical Environmental Assessment' software (GEO-EAS, USEPA 1988). Geochemical modelling and speciation determination was done with an updated version of the PHREEQE software (Plummer, Parkhurst & Thorstenson, 1980).

W1/28.10	Zn		Cu		Pb		Cd							
	μg/l													
VDAL LOD C	24		22		2		~1							
AKAL ICP-S	24		23		2		<1 < 1							
I UD ULAA	25		<i>42</i>		J		~ .1							
W1/30.3.	Fe		Mn		Zn		Cu		Pb		Cd			
	μg/l													
									_		_			
XRAL ICP-S	126		167		8		4		nd		nd			
FUB GFAA	nd		nd		10		2		<1		<.I			
OTML GFAA	102		76		8		3		<1		0.05			
W3/1.4.	Fe		Mn		Zn		Cu		Pb		Cd			
	μg/l													
XRAL ICP-S	111		21		28		<2		nd		nd			
FUB GFAA	nd		nd		20		1.5		<1		<.1			
OTML GFAA	87		24		19		1.3		<1		<.04			
W1/7.4.	Fe		Mn		Zn		Си		Pb		Cd			
· · ·	μ <u>g</u> /l												 	
XRAL ICP-S	319		13		17		53		nd		nd			
FUB GFAA	nd		nd		20		50		<1		<.1			
OTML GFAA	134		15		15		34		<1		0.1			
W3/8.4.	Fe		Mn		Zn		Cu		Pb		Cd			
	 μg/l								- ~					
	10													
XRAL ICP-S	50		<5		<5		14		nd		nd			
FUB GFAA	nd		nd		<5		16		<1		<.1			
OTML GFAA	48		4		1.3		16		<1		0.08			
Blank (DDW lab wate	er)	Fe		Mn		Zn		Cu		Pb		Cd		
		µg/l												
XRAL ICP-S		<10		<5		11		<2		nd		nd		
FUB GFAA		nđ		nd		20		<1		<1		<.1		
OTML GFAA		27		0.5		13		0.2		<1		<.04		

Table 2. Quality control for water samples. The same sample was analyzed by three different laboratories.

3. THE ENVIRONMENT OF THE OK TEDI/FLY RIVER REGION

3.1. Geology of the Mount Fubilan Ore Deposit

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Trace metal contamination of the Ok Tedi/Fly River system through mine discharges is closely related to the geochemical composition of the Ok Tedi ore.

The orebody at Mount Fubilan consists largely of an intruded and altered monzonite porphyry stock of Lower Pleistocene age, which hosts a mesothermal, stockwork and disseminated copper-gold mineralization (Rush and Seegers 1990). Intense weathering at Mt. Fubilan led to the formation of a copper-depleted, but gold-enriched cap (gossan) overlying the primary mineralization. The deposit is similar to other big copper porphyry mines in the Pacific Rim region. Pyrite and chalcopyrite are the dominant sulfide minerals in the protore below the supergene enrichment zone. Average metal concentrations in the orebody are 0.75% of copper and 0.67 grams per ton of gold (OTML 1993). At the contact between the intrusive and adjacent sediments, calc-silicate, sulfide and magnetite skarns have formed, which make up approximately 10 % by volume of the orebody (Jones and Maconochie 1990).

Porphyry base-metal deposits of hydrothermal origin like Ok Tedi typically contain a paragenetic sequence of the metals copper, molybdenum, silver, gold, lead, zinc, iron, manganese, selenium, arsenic, cadmium, tungsten and others, which are not evenly distributed within the orebody, but are located in distinct zones around the low-grade quartz core of the intrusion. Very high values for lead (1000-3000 ppm) have been found in overburden sampled in October 1991. The skarn mineral paragenesis, has an elevated and more variable content of trace metals as compared to the copper

porphyry ore, and is more critical from an environmental point of view.

However, arsenic and mercury, which are frequently associated with gold mineralisation, show no enrichment above background levels in the Ok Tedi deposit.

Ore minerals of the oxide zone: cupriferous goethite Cu.FeOOH, cuprite Cu₂O, native Cu, malachite Cu₂CO₃(OH)₂, azurite Cu₃(CO₃)₂(OH)₂, copper carbonates

Ore minerals of the sulfide zone: chalcocite Cu_2S , digenite, covellite CuS (supergene formations) chalcopyrite $CuFeS_2$, bornite Cu_5FeS_4 (very minor), pyrite/marcasite FeS_2 , molybdenite MoS_2 (protore mineralization)

Sulfides of Se and Ag are accessory minerals in the main orebody (Fubilan Monzonite Porphyry)

Pb, Zn, Cd, As, Ag and Se minerals are found mainly in skarn ores

 Table 3. Ok Tedi ore composition

3.2. Geomorphology and Vegetation

The minesite at Mount Fubilan, which had an elevation of 2094 m above mean sea level (MSL) before mining started, is located in the upper catchment of the Ok Tedi (Map 1). The site receives rainfall of 10,000 mm per annum, which is amongst the world's highest.

Dense tropical rainforest blankets the ridge and ravine topography. The area is seismically active and prone to landslides. North of the Ok Tedi catchment stretches the massive Hindenburg Range with maximum elevations of 3325 m. The range is a highly unstable, cliff-like structure build up of Tertiary Darai Limestone, which is responsible for the high natural calcium and bicarbonate content in Ok Tedi water.

The tributary streams in the area are very steep and form narrow, gorge-like valleys, with boulder-size alluvial debris. Upstream of Ningerum (60 m above MSL), the Ok Tedi is a braided river, up to 100 m wide. At Ningerum, 70 km to the south of the mine, the Ok Tedi River leaves the mountaineous region. Between Ningerum and Konkonda, the river passes through a transition phase and enters the Fly Platform, a vast alluvial plateau with very little relief and meandering streams. 130 km downstream of Ningerum, the Ok Tedi flows into the Upper Fly River at D'Albertis Junction at 20 m elevation above MSL. South of the junction, the river channel is flat and meandering, flanked by thick jungle, extensive backswamps, lakes and lagoons. The reach between D'Albertis Junction and the confluence with the Strick-land River south of Obo (Everill Junction) is called Middle Fly River, with a length of 410 river kilometers. The sinuosity factor is about 2.4. The channel width varies between 200 and 300 m on average.

Downstream of Kai and Agu Rivers/Lakes, the floodplain vegetation pattern is dominated by tall, dense reed (*Phragmites, Saccharum*) and smaller grasses. Jungle is restricted to a few isolated areas of higher elevation. Trees grow only in rarely inundated areas (with the exception of Sago palms and thin-stemmed *Melaleuca*) because of their susceptibility to prolonged flooding. The fact that trees are largely missing in the lower part of the Middle Fly points to frequent natural flooding.

Lakes and shallow water bodies and channels of the floodplain often have a dense vegetation of aquatic grasses, floating *Azolla* (Equisetum), *Pistia* (Araceae), water lillies, and the submerged *Ceratophyllum*.

Below Everill Junction, the Lower Fly River is tidally influenced, reaching the delta 400 km downstream. Due to the increased flow rate and a lower gradient, the channel width and meander amplitude are greater than in the Middle Fly.

The Fly River floodplain is made up of an inner part, the channel migration zone comprising the present river channel, cut-off meanders and meander scroll complexes, and the outer part with drowned valley lakes and extensive backswamps (Fig. P1).



Fig. P1. Backswamp close to Obo, several hundred metres away from the Fly River. Due to difficult access, only few samples could be taken from these locations.

The floodplain widens from 4 km width near Kiunga to over 14 km close to Obo. Discontinuous low natural levees, about 1-2 m above mean river level, flank the existing river channel.

Drowned valley lakes are typically connected to the river by a narrow, sinuous channel with a length of 1-2 km, 3-10 m width and a depth of 2 to 5 m, becoming more shallow towards the lake (Fig. P2). The channel itself is being kept open by the scouring action of sediment particles in the water. The lakes have a depth of about 1.5 to 2.5 m, cut by slightly deeper channels, decreasing to less than 1 m towards the end away from the river. Their bottom relief clearly reflects their origin as tributaries to the Fly River. During the sea level rise which ended about 5,000 years ago, the river ends of the larger floodplain creeks were filled with sediment derived from the Fly River. Water can flow freely in both directions, depending on the water levels in lakes and the Fly River.

Due to the low relief, the edges of lakes or "lagoons" are not well defined and tend to merge with neighbouring grasslands at high water level (Fig. P3). Higher banks, on which the villages are usually located, are remnants of eroded plateaus of Pleistocene age. The water level in the lakes is highly variable. Following consecutive dry months, lakes may dry out completely which can result in massive fish kills.

3.3. Hydrology and Climate

The climate in the investigated area is wet tropical. The minesite receives well distributed rainfall on 325 days throughout the year. Average temperatures show a nocturnal minimum of 20°C and a mean daily maximum of 27°C.

Mean annual relative humidity is about 85% in the Fly River lowland. Temperatures are constant throughout the region with an annual mean of 29°C.

The Fly River has a total length of 1120 km. Its most important tributaries are the Ok Tedi and the Strickland River. The catchment comprises more than $76,000 \text{ km}^2$. The mean annual runoff per unit of catchment area is about 2500 mm, which is higher than for any other river system in the world (Maunsell and Partners 1982).



Map 1. The Fly River Region.

As the river catchment is relatively small, streamflow is characterized by short-term water level fluctuations with rapid changes between flood peaks and drought periods. Rainfall is highest in the upper Ok Tedi region, with recorded annual values up to 14,000 mm, and 7,800 mm at Tabubil, and decreases in the lowland. At Kiunga on the Upper Fly River the annual mean is 4,700 mm, at Lake Bosset in the Middle Fly area 2,670 mm, and at the coast at Daru 2,100 mm. Both rainfall and river flow show only limited seasonality. June to October usually are the dryest months in the Middle Fly region.



Fig. P2. The channel connecting Lake Bosset with the Fly River (right background). Deposition of riverine material is highest along the channel.



Fig. P3. The Fly River, inundated floodplain swamps and lakes, and jungle on the alluvial plateau in the background.

The long-term mean discharge rate of the Ok Tedi at the junction with the Upper Fly River (flow 1178 m³/s at Kiunga) has been measured at 923 m³/s. The combined flow is 2161 m³/s in the upper Middle Fly. There exists a water exchange between the floodplain and its water bodies and the Middle Fly River. The net supply of sediment-poor, but DOC-rich floodplain water to the Middle Fly, however, appears to be small. At Obo, the long-term mean discharge is 2244 m^3 /s. The Strickland River discharges 3110 m^3 /s into the Fly River at Everill Junction (OTML 1994). Mean daily flow data for the last few years are summarized in Table 4.

Water level fluctuations in the Middle Fly are smaller than in the Upper Fly River, where water level changes of up to 15 m have been recorded at Kiunga. The discharge of the Fly River at its mouth of 6.000 m³/s makes it comparable in size with the Niger and Zambesi in Africa or the Danube in Europe.

	1988	1989	1990	1991/92	1992/93
Di Lita	*	25	27	24	27
Bukrumdaing	Ŧ	25	27	24	27
Tabubil	175	110	*	*	*
Ningerum	305	222	183	193	290
Konkonda	965	850	1122	456	805
Kiunga	1435	1197	1044	807	941
Kuambit	2400	2124	2094	1407	2061
Obo	2515	2613	2424	2057	1 978
Strickland	3785	3869	3769	2870	3141
Ogwa	6300	5792	5461	*	*

Flow rate (m³/s)

* no data available

Table 4. Mean daily flow data collected by OTML for the Ok Tedi, Fly and Strickland Rivers at different locations in the catchment.

3.4. Aquatic Biology

A study conducted prior to OTML's operation (Roberts 1978) came to the conclusion that the Fly River system supports the most diverse fish fauna in the Australasian region, with at least 105 freshwater species from 33 families. The Sepik River in Northern New Guinea, the river closest to the size of the Fly on the island, has a relatively low overall fish density with a total of 57 freshwater species (Allen & Coates 1990).

The fish population in the Fly is remarkable for the large size of some species (e.g. black bass and barramundi) and the abundance of endemic species like catfish. The most targetted species by commercial fisheries is the barramundi *Lates calcarifer*, which roams the entire length of the Fly and resides in floodplain water bodies during most of its lifecycle. The barramundi migrates annually to the coastal areas for spawning (Eagle 1993).

Fish ecology is characterized by an overlap of species types resident in the two main habitats, the river channel and the floodplain with its waterbodies. Due to the high biological productivity of the Fly floodplain, the majority of the food for the Fly River fishes originates from off-river sources (Kare 1992). Overlap in diet and habitat requirements is an important mechanism for survival since prolonged periods of low water level may result in drying out of the floodplain and its shallow water bodies (Eagle 1993). Under these conditions, the fish take refuge to the stream channel and oxbow lakes, which may represent the only standing water on the floodplain. The fish populations decline substantially, however, the recovery usually is rapid with recolonisation of the newly flooded habitat by the surviving stocks (Smith & Bakowa 1994). The most important food items for fish are aquatic and terrestrial invertebrates (freshwater prawns, mayfly larvae and other insects, worms etc.), algae, plant and organic detritus. Predatory fish like barramundi and catfish feed on smaller fish like *Nematalosa* herrings.

3.5. Population

In the entire Ok Tedi/Fly River drainage area lived in 1980, according to a national census, about 73,500 people. They speak 28 different languages and form five different language families: The Ok and Awin people of the Ok Tedi and Upper Fly River, the Marind of the Middle Fly and Lower Strickland (Lake Murray) region, the Suki-Gogodala people of the Lower Fly, and the Trans-Fly peoples of the southern coastal plain. The mountain people are hunter-horticultur-alists living on a subsistence economy. Fish becomes a more important part of the diet for the lowland riverine people. In the Middle Fly-Lake Murray region, habitable land and ground suitable for cultivation of plant foods is scarce. The inhabitants (about 4,500 in 1980) are mainly hunter-gatherers who collect wild sago, cooking-bananas and sugar cane, and use the fish resource of barramundi, black bass and catfish (Busse 1991). The sale of crocodile skins is an important source of cash. The largest populations in the Middle Fly area are at Lake Bosset, Lake Pangua and Lake Daviambu. Small villages and homesteads are typically located along lakes and rivers. Virtually all travelling is by dugout canoe. During extremely dry periods, in which lakes may dry out completely, people abandon their villages and move to the main rivers which remain the only source of water.

The development of the Ok Tedi Mine has resulted in massive cultural and socioeconomic changes in the affected region. A former government adviser emphasized the "psychological trauma that Ok Tedi development will bring to the simple life styles of the mountain people" (Pintz 1984).

The mining company has established the "Lower Ok Tedi/Fly River Development Trust" which offers basic village infrastructure in the field of transport, health, education and business development. It supports an estimated 30,000 people in 101 villages living along the river system, who may be negatively affected by the discharge of mine residues and associated problems.

4. INPUT OF THE OK TEDI MINE INTO THE OK TEDI/FLY RIVER SYSTEM

4.1. Description of the Mining Project

Mining of the Mount Fubilan orebody as an open pit started in 1984, sixteen years after the discovery of the deposit by Kennecott exploration geologists in 1968. The particular features of the ore deposit were responsible for the development of the mine in mainly two stages: from May 1984 to late 1988, the gold-enriched cap was mined and processed. This involved the use of sodium cyanide and other process chemicals for gold extraction. Following this stage, the phase of extracting copper ore commenced. The gold is now recovered in the sulfide flotation concentrate. Mining of the Mount Fubilan deposit will continue until 2008 and involves the handling of some 1,400 million tonnes of material, comprising 550 Mt of milled ore residues and 850 Mt of overburden (Salomons & Eagle 1990).

Currently, the production is about 80,000 tons of ore and a similar volume of waste rock per day. The latter is material in which the metal content falls below the cut-off grade of 0.2% for copper or 0.8 g/t of gold.

The currently produced ore has 0.84% Cu and 0.71 ppm Au (oral comm., OTML, 1994). The sulphide flotation process has a recovery rate of 85% which leaves about 1,300 ppm Cu in the residual material. The copper content of waste rock has an average of 1,100 ppm (OTML 1994).

The ore is crushed and subsequently fed to a series of ball mills where the material is ground to fines. A copper-rich concentrate, containing all valuable metals, is recovered by means of a conventional sulfide froth-flotation process. For maximum sulfide recovery and pyrite depression, the pH of the flotation mixture is adjusted with lime to 10,5-11 and organic process chemicals are added (England et al. 1991).

The final concentrate is transported by a 160 km slurry pipeline to the river port of Kiunga on the Upper Fly River. At the Kiunga wharf, the slurry is filtered, dried and stored to await shipment by bulk carriers down the Fly River. From a storage facility in the Fly River Delta, the concentrate is sold under long-term contracts to smelters in Japan, Germany, South Korea, Finland and the Philippines.

In 1994, OTML produced 206,329 tons of copper in concentrates, containing also 15.14 tons of gold, 30.20 tons of silver, and other metals. The mine is the world's fifth biggest copper producer. Ok Tedi's mineable reserves in early 1994 were estimated at 459 Mt with 0.72% Cu and 0.70 g/t Au (DMP 1994), sufficient for another 14 years of operation.

Following a recent restructuring of ownership, the shareholders of Ok Tedi Mining Limited are as follows: 52% of shares are held by Broken Hill Proprietary, the biggest Australian mining company, which is also the project's operator; the Government of Papua New Guinea has increased its stake from 20 to 30%, and 18% of shares are now being held by Inmet of Canada (formerly Metall Mining, a German-Canadian mining company) (Mining Journal, August 18, 1995).

4.2. Discharge of Tailings and Waste Rock

The tailings from the copper extraction in the mill, approximately 98% of the original feed to the processing plant, is piped without further treatment to the Ok Mani, a tributary of the Ok Tedi (Fig. P4). Waste rock is hauled to erodible dumps adjacent to Mt. Fubilan, from where the material is washed into the headwaters of the Ok Tedi.



Fig. P4. The narrow valley of the Ok Mani, which drains the Mt. Fubilan area, has been filled up by some 20 m, as can be seen at the drowned rainforest trees.

The thickened, alkaline tailings slurry consists of approximately 55% solids of which 78% have a grain size less than 100 μ m. The material has about 10-20% of its original copper content, varying amounts of trace metals including zinc, lead and cadmium which occur naturally in the porphyry ore, and small quantities of organic flotation chemicals.

Waste rock and overburden is coarser, but has a high percentage of soft material (siltstone and limestone) which breaks down easily, containing copper and significant quantities of other heavy metals. Although some of the waste rock remains temporarily in the dumps and adjacent creek valleys (depending on the rainfall activity), most of it is washed down rapidly into the Ok Tedi River.

Since the beginning of mining at Mt. Fubilan, all waste rock and overburden (with the exception of a short period during the gold stage) has been disposed off in the river system. The construction of a tailings dam was halted in early 1984 when a land slide forced the abandonment of the construction site, which was located in a geotechnically highly unstable zone.

The larger part of the mine-derived material entering the upper Ok Tedi has a particle size of less than 100 µm and can be transported as suspended load throughout the entire length of the Ok Tedi/Fly River system, given sufficiently high hydraulic transport capacity.

The massive input of mine-derived sediments into the Ok Tedi exceeds the sediment transport capacity of the river system, which has led to severe aggradation of the river and a rise of the Ok Tedi channel bed by ten meters and more in the upper reaches. Riverbank food gardens and plantations have been flooded and the natural ecology of the river and subsistence fishing has been seriously disrupted.

About 60 million tons of material are delivered to the Ok Tedi/Fly River system per year, containing approximately 69,000 tons of copper. The daily discharge rates are 160,000 tons of rock material with 190 tons of copper.

The presence of iron and base metal sulfides in the waste rock and tailings indicates the possibility of acid mine drainage (AMD) development, which arises from the oxidation of sulfide minerals and subsequent sulfuric acid production. AMD generation in the Ok Tedi waste rock dumps was considered a potentially serious problem in the Ok Tedi Environmental Study by Maunsell and Partners (1982). Low pH in waste rock drainage may result in enhanced solubility of trace metals like lead, silver, zinc, cadmium and copper (Ferguson and Erickson, 1988). However, acid formation may be buffered by alkalinity released from carbonate minerals in the mine waste, such as calcite (CaCO₃). Given the relatively high calcium content in the waste material, development of AMD in the waste rock dumps appears unlikely. Despite the fact that sulfide oxidation is occurring in the mine discharges, no pH values below 7 have been recorded in water of the upper Ok Tedi, neither by OTML nor in measurements during the present study.

5. SEDIMENTOLOGY OF THE FLY RIVER AND ITS FLOODPLAIN

5.1. Natural Sedimentary Processes in the Floodplain

The Fly River has a highly variable flow regime where extensive flooding and overbank deposition on its wide floodplain alternates with extremely low water levels in drought periods. The sedimentary processes in the Fly River floodplain are dominated by the river itself and the suspended load it carries. Sediment delivery by floodplain creeks and channels to the system is negligable. The catchment in the outer floodplain and the adjacent piedmont plain comprises forested areas with some swamp savannah. The content of particulate matter in floodplain creek waters generally is very low (below 15 mg/l), consisting of organic material, mainly plant debris, and strongly weathered soil material.

At mean flow in the Fly River and low water table in the floodplain, intrusion of riverine suspended sediment occurs along pre-existing channels linking the river with the floodplain. Sedimentation is highest in oxbow lakes and tie channels, where medium to coarse silt from the river suspended load settles down, and much lower in drowned valley lakes and floodplain depressions where clay and fine silt may become deposited.

Drainage direction throughout most of the Fly River floodplain is towards the Fly, although the inner floodplain, the active meander belt, tends to be higher in elevation than the outer floodplain due to sedimentation along and in the main river channel. No broad continuous levee is developed along the river and where there is a dam, it is cut by small drainage channels. During periods of high water level in the Fly River, which are associated with higher sediment loads because of increased transport energy, river bank overflow and inundation of the alluvial plain occurs. Overbank flooding during moderate floods is localized close to the Fly River channel in flanking swamps. The water returns to the river when the flood recedes, although most of the suspended load carried into the floodplain will be deposited there due to vegetative filtering in the grassland bordering the main river, and in sediment traps such as lakes.

Maunsell and Partners (1982) recorded a period of very high flow in the Fly River in July 1981, which was the highest since 1977. The water level was about 1 m above the natural levees. According to their observation, grassland and forests adjacent to the river channel in the reach between Lake Bosset and Obo were flooded to a width of about 16 km on either side of the channel. Although this situation is highly anomalous, it is evident from aerial photographs that at high water level in the Fly River, turbid flood water may flow several kilometers across the floodplain.

The background deposition rate (i.e. before the mine started discharging material) in drowned valley lakes is very low. Analysis of the fraction less than 2 μ m of sediments taken from small islands (0-1 m above mean lake level) in Lake Bosset and Bai Lagoon (Map 2) defined a mineral composition dominated by kaolinite, aluminium chlorite, quartz and gibbsite. All four minerals are typical of highly weathered tropical sediments. The fact that sediment of probable Pleistocene age was found on the surface of these shallow islands indicates that no Fly River material has been deposited there in the last 120,000 years, when the sea level began to drop and Pleistocene sedimentation in the alluvial plain ended.



Map 2. The lower part of the Middle Fly River (study area).

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Core	3/30.3.	3/11.4.	1/5.4.	1/9.4,	
Locality	L. Daviambu W	L. Bosset NW	Bai L. C	Kai L. SW	
Depth of peaty layer below sediment surface	13 cm	28 cm	27 cm	28 cm	
C14 age (years before present)	5265 ±215	2950 ±70	4030 ±110	4380 ±125	
Sedimentation rate mm/1000 years	24	95	67	64	
Water depth at site (mean water level)	1.10	2.40	1.70	2.60	
Distance from Fly (air km)	6	5	3	9	

Table 5. Results of radiocarbon age dating of sediment cores from drowned valley lakes.

Table 5 shows the results of C14 age dating which was performed to determine sedimentation rates in drowned valley lakes. The overall deposition rate seems to be well below 1 mm/year and may be as low as 0.1 mm/year. The sediment deposition at a particular site within a lake is very much dependent on bottom relief and shape of the lake. Core 3/30.3, was taken from the distal end of Lake Daviambu. The upper 13 cm consist entirely of peat-like organic debris. Accumulation of this material, derived from the catchment of the lake itself, is much slower than at those sites which receive suspended sediment from the Fly River. In the case of the other three cores, the sediment layer above the peat was made up of Fly River material, as determined from its characteristic clay mineral assemblage with dominantly (low charged) smectite. A thin layer of copper-rich material on the top of the cores was observed.

Due to the lack of dateable organic matter in sediment cores from tie channels and their banks, where sedimentation certainly was much higher, no data are available to establish deposition rates at these sites. The simple fact that the shallow lakes still exist after about 5,000 years of possible sedimentation from the Fly River points to very low deposition rates. Assuming that the maximum channel depth of 5.4 meters, measured in Lake Bosset channel close to the Fly River, reflects the original channel depth of the drowned tributary, then the deposition rate was about 1 mm/year in the last 5,000 years at sites close to the Fly.

At times when discharge from the Strickland River is greater than the flow of the Fly, a backwater effect develops, which results in decreasing river velocity in the Middle Fly and subsequent settling of suspended material on the river bed. The same occurs during low-flow periods. The deposited material may be resuspended at high flows. Pickup et al. (1979) estimated the natural suspended load of the Middle Fly at 7-10 million tons and bed load at 1-2 million tons per year.

5.2. Deposition of Mine-Derived Material

The suspended sediment concentrations in the Middle Fly River today are 5 - 10 times above the natural load of about 60 mg/l. Significant quantitities of mine-derived sediments are deposited and trapped in creeks, lakes and swamps adjacent to the Fly River. Such off-river sites play an important role in the food web and in the reproduction cycle of aquatic organisms like invertebrates and fish.

Flow inversion in channels linking the floodplain with the Fly River is an important process because it is responsible for suspended sediment transport to off-river sites during mean flow conditions (Fig. P5-6). Reverse flow upstream the tributary channels was frequently observed during the three field trips undertaken. On April 8, 1993 slightly turbid water (sample W1/8.4.) with elevated calcium, copper (9 μ g/l) and cadmium (0.12 μ g/l) levels was encountered in the

Agu River/Lake system about 25 km upstream of the junction with the Fly River, which had about mean flow. The Agu River/Lake in this reach runs roughly parallel to the Fly. As a result of a recent Fly River intrusion upstream the Agu River, turbid water was visible in the densely vegetated parts of Agu Lake, whereas in the Agu main channel, Fly water was slowly pushed out in southerly direction (downstream) due to heavy rainfall in the days before the site was visited. A similar observation was made in the Kai River/Lake system which was sampled on April 9, 1993, although in this case Fly River water was still flowing upstream. The upstream current observed close to the Kai River mouth was remarkably strong and made the use of the outboard motor necessary for passage.



Fig. P5. The channel linking Lake Daviambu (background) with the Fly River. Lake water of light orange-brown colour, rich in humic substances, drains into the Fly. The vegetated areas in the foreground are swamp.



Fig. P6. The same location, but at a higher water level in the Fly River. Turbid, white river water flows into the lake.

Water with high conductivity, neutral pH and elevated copper values (39 μ g/l in unfiltered sample W4/9.4.) was encountered 19 channel km upstream of the Kai River confluence with the Fly River. Water movement at the sampling location was in northerly direction (upstream). One could expect that there exists a flow-through mechanism in which Fly River water enters the Kai River/Lake system (which is about parallel to the Fly) through an upstream connection.



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However, satellite images and aerial photographs show no such connecting channel and give no indication of flow across the floodplain, away from the river. The same observation was made for the Agu River system, although maps show a northerly channel connection of the Agu with the Fly River. According to local villagers, this flow-through mechanism is only active at very high water level in the Fly.

The frequent intrusions of highly turbid Fly River water should result in widespread sedimentation of mine-derived material in the Kai and Agu Lakes. Sediment sample analysis revealed moderate copper pollution. Only the uppermost 3-5 cm of sediment cores showed strongly elevated trace metal contents with Cu around 300-600 ppm. In the Kai River samples, the contaminated sections consisted of deposited riverine suspended matter. In the Agu River system, the uppermost, copper-rich sediments were mainly made up of humic material in which the high copper content may be secondary, due to adsorption on flocs of organic matter.

Deposition of mine-derived sediments generally is highest at locations close to the channels which connect floodplain water bodies with the Fly River. Comparatively coarse material (medium to coarse silt) is deposited there, whereas the remaining fractions of clay and fine silt of the riverine suspended load is carried into the floodplain water courses (Fig. S1-S4). Trace metals are generally enriched in the finest particle fractions.

In a sediment core from the Kai River channel bank (1/10.4.), taken approximately 3 km upstream of the Fly River confluence, the upper 38 cm showed strongly elevated copper values (400-700 ppm). This material must have been deposited within the last ten years, since the Ok Tedi mine is operating. The sedimentation rate is about 4 cm/year. Similar observations were made at the other drowned valley lakes investigated (Bosset, Kongun, Kemea, Bai, Daviambu). Where the well defined channels extend into the lake, substantial deposition of 20-40 cm of copper-contaminated sediment was detected several kilometers away from the Fly River, whereas in the proper lake copper-rich sedimentation did not exceed 5 cm. Sediment core 4/11.4., which showed elevated trace metal levels in the upper 23 cm with about 200 ppm copper at the bottom and 800 ppm at the top section, was taken in Lake Bosset at the location where channel and lake water merge, about 2 km inside the lake (as depicted in topographic maps). Sedimentation of mine-derived material was highest in oxbow lakes with up to 70 cm measured at sites close to the river end of Lake Pangua and Lake Kibuz (Fig. P7). 29 oxbow lakes at different infilling stages exist along the Middle Fly River.



Fig. P7. Inflow of Fly River water, rich in mine-derived suspended matter, into the Pangua oxbow lake. Deposition of copper-rich material is highest at the river end of the lake.

Two sediment cores were taken from the Fly River channel bed (Fig. S5-S7), one at Lake Pangua (water depth 10.5 m) and the other downstream of Obo (depth 14.5 m), both in the deepest part in the channel cross section. The entire bed core from the Lake Pangua site (56 cm) consisted of fine grained sediment with 55-75% finer than 20 μ m (clay, fine and medium silt fraction), and displayed a constantly high copper content of about 850 ppm.



The upper 30-35 cm of the core taken close to Obo were very similar in composition, however the bottom section (41-43 cm) showed much coarser material with 75% fine sand (63-200 μ m). The copper content dropped to 48 ppm in this section. From both cores, it is evident that suspended load has settled down on the channel floor. Only the section with mainly fine sand, from the sample taken downstream of Obo, shows the typical grain size of bed load. It is known from earlier sedimentological studies (Pickup et al. 1979) that fine material does temporarily deposit on the river bed upstream of Everill Junction. Due to the cohesive forces among fine grained particles, high current speeds are necessary to erode these deposits. The resuspension of mine-derived clay- and silt-size material from the channel floor today may be limited because much more sediment settles down, and will be flushed downstream at high flows only at hydraulically preferred sites like meander bends and immediately upstream of the Strickland junction.

In this connection, it is interesting to note that two islands within the Fly River channel, one immediately downstream of Lake Bosset and the other at the junction of Tamu Creek, have been observed during the field trips undertaken. Sediment samples taken showed relatively coarse, copper-rich material (Fig. S8). No such islands are visible in aerial photographs from the 1960's and 1970's. The formation of islands in the river channel also indicates insufficient transport capacity of the Fly River to carry all mine-imposed waste material. OTML (1993) reports an increase in bed level at Kuambit, immediately downstream of the Ok Tedi/Fly River confluence, of more than one meter above the 1984 premine baseline.

Two effects probably are responsible for the strongly increased deposition rates as compared to earlier sedimentation: Firstly, because of the increase in the (mine-derived) suspended sediment load of the Fly River, which today also contains more silt than the natural particulate load. Secondly, as a result of the reduction in channel capacity due to bed aggradation, overbank flow frequency increases. In the lower part of the Middle Fly, close to the Strickland River junction, aggradation of the main river channel is higher due to the backwater effect from the Strickland.

Sedimentation studies which were undertaken before the Ok Tedi mine was developed (Pickup et al. 1979) assumed that frequent landslides, particularly in the upper Ok Tedi catchment, have lead to a high natural suspended sediment load in the Fly River. Had this been the case, this should have resulted in much higher natural deposition rates in the Middle Fly floodplain. The comparison of sedimentation rates before and after the mining at Mt. Fubilan started, points to the fact that the background suspended sediment concentration in the Fly River was very low, probably in the range of 40-60 mg/l.

		ss/gulp (mg/l)	ss/D)	l (mg/l)	
	1988	9/88-8/90	10/91-9/92	1990/91	1991/92	
Bukrumdaing	*	151(3)	44(5)	57	64	
Tabubil	4709	7825(25)	6696(12)	8095	10626	
Ningerum	1802	524(124)	3477(12)	5846	5882	
Konkonda	698	1420(24)	949(12)	2072	2744	
Kiunga	*	60(24)	26(12)	186	149	
Kuambit	305	520(24)	316(12)	625	573	
Bosset	*	171(23)	118(12)	*	*	
Obo	65	92(25)	103(12)	649	450	
Strickland	*	414(24)	239(12)	690	509	
Ogwa	326	283(23)	*	677	323	

* no data available

Table 6. Average data on suspended sediment concentrations in gulp (surface) and depth-integrated (DI) samples. Medians shown and number of measurements in brackets. Data from OTML (1988-92).

Attempts were made to establish a suspended sediment balance using data collected by OTML. A proper balance would show sediment losses (or gains) in the system. A brief examination of data in Table 6 shows that there is a large scatter in the measured concentrations over several sampling periods. Of main interest to the present study is the deposition of suspended riverine material in the Middle Fly region. The measured suspended matter content in gulp samples of near-

surface water in the Fly River reach between Kuambit and Obo shows a strong decrease in all sampling periods. This effect can not be seen clearly in the depth-integrated measurements in which the suspended matter content is determined in vertical sections through the water column.

Typically, the suspended sediment concentration of near-surface river water (gulp samples) is 70-90% of the depthintegrated suspended matter content (Higgins 1990). No such relationship can be established in the data of Table 6. Differences between suspended sediment content determined by gulp and depth-integrated sampling were up to 90% in measurements from the same site and date (data in OTML 1993). It is difficult to explain this discrepancy, which is particularly evident at Obo. There obviously exists a pronounced stratification within the flowing water body, with particulate-rich water travelling close to the bottom. Due to the shift of the main current in river bends, which leads to a more turbulent (spiral) flow, the bottom strata with a higher suspended solids content rise to the surface. This effect was observed in the depth-integrated measurements at Obo where suspended sediment concentrations are usually highest close to the outer bend.

The stratification in suspended sediment concentration within the water body points to the settling down of suspended load on the channel floor. River bed sedimentation is largely dependent on flow velocity, which at Obo is significantly lower than at Kuambit/Nukumba. A detailed comparison of flow data gathered in February and April of 1992 by OTML gave a mean value of 1.12 m/s for Kuambit and 0.66 m/s flow velocity at Obo.

Ok Tedi Mining Ltd. has tried to establish an annual suspended sediment mass balance based on its regular measurements taken in the Ok Tedi and the Fly River (Table 7). The calculations illustrate the complexicity of the system and the difficulties in predicting sediment transport in the two rivers. The measured load values (calculated from a small number of measurements) in the 1991/92 data indicate a loss of sediment from Ningerum to Obo (46 to 35 Mt/a), whilst the model predicts a steady increase in the same reach (43 to 55 Mt/a). In the data set for the following year, the discrepancy between observed and predicted load is minor. Given the fact that sediment input by the Ok Tedi mine into the system is fairly constant over the years, the massive adjustment of the predictive model between the two years (at Obo 38 instead of 55 Mt/a) is hard to understand. The flow weighted load data for 1992/93 suggest heavy erosion of material in the Middle Fly between Kuambit and Obo (increase of suspended load by 4 Mt/a), which is hard to explain, too. This may be attributed to the limited number of data collected (9 depth-integrated samples).

Station	Flow Weighted	Flow Weighted	Predicted	Predicted
	Load (Mt/a)	Conc. (mg/l)	Load (Mt/a)	Conc. (mg/l)
Nindorum	16	5750	42	5001
Konkonda	40	2001	43	1860
Kuambit	40 37	646	45	724
Obo	35*	450 [*]	55	687
Strickland R.	84*	900*	85	-
Ogwa	119*	323*	140*	807
Station	Flow Weighted Load	Predicted I	Load	
	(Mt/a)	(Mt/a)		
Ningerum	40	-		
Konkonda	31	35		
Nukumba/	34	38		
Kuambit				
Оbo	38	38		

* insufficient records; estimates based on available data

Table 7. Suspended sediment mass balances for 1991/92 (above) and 1992/93 (below) calculated by OTML (1993, 1994).

6. POTENTIAL MOBILIZATION OF HEAVY METALS AND HYDROCHEMISTRY

6.1. Heavy Metals in Sediments of the Ok Tedi/Fly River System

Analytical results for sediments from the Fly River floodplain are shown in Tables 8 and 10 (A1 and A3 in annex) and from the upper Ok Tedi in Table 9 (A2 in annex). The samples from the Fly River section are grouped into two populations, i.e. sediments which were deposited before the Ok Tedi mine started discharging residues into the Ok Tedi/Fly River system (Table 8), and sediments controlled by mine-derived material (Table 10). This classification is easily practicable due to the distinct geochemcial signature in the mine-derived material, which shows for some elements a significant enrichment above the natural background. The frequency distribution plotted in log probability graphs for copper in 385 sediment samples is shown in Fig. D1 and for gold (27 samples) in Fig. D2. The distributions are roughly bimodal. The subpopulations have approximately lognormal distribution.

Parameter	Mean	SD	Median	Percen	tiles	
				25	75	
					·	
Na mg/kg	4365	3146	3700	2000	5400	
K mg/kg	10829	4448	12050	7200	14100	
Mg mg/kg	5335	2320	5900	3400	7200	
Ca mg/kg	8787	7493	6100	4875	8025	
Al mg/kg	85538	25348	91100	71700	102400	
Fe mg/kg	37302	15185	33700	7200	46300	
Mn mg/kg	289	225	197	136	334	
Zn mg/kg	142	54	130	115	164	
Cu mg/kg	45	19	44	32	54	
Pb mg/kg	18	9	18	11	23	
Cd mg/kg	0.26	0.13	0.2	0.2	0.2	
Au μg/kg	6.7	6.5	3.0	1.0	8.5	
Ag mg/kg	0.24	0.26	0.2	0.1	0.3	
As mg/kg	4.6	3.6	4.0	2.0	5.0	
Cr mg/kg	61	16	61	55	70	
Mo mg/kg	1.7	1.4	1.0	1.0	2.0	
Co mg/kg	13	6	12	9	16	
Ni mg/kg	33	13	29	24	41	
V mg/kg	165	49	171	143	196	
Ti mg/kg	3765	1269	4100	3100	4400	
Zr mg/kg	67	25	67	49	83	
La mg/kg	24	7	25	20	28	
Ba mg/kg	309	119	309	255	361	
Sr mg/kg	137	57	127	101	162	
Sc mg/kg	17	5	18	14	20	
С%	6.7	9.4	2.6	1.3	6.0	
S mg/kg	1447	1445	800	300	2525	

Table 8. Mean, standard deviation, median values and 25% - 75% confidence intervals for 27 elements in Middle Fly River background sediment samples (n = 128).

Gold gives the highest enrichment factor of 53, followed by molybdenum (factor 23), copper (factor 12), lead (factor 4.4), calcium (factor 4.1), silver (factor 3.5), sulfur (factor 2.6) and strontium (factor 2.5). Zinc shows an enrichment factor of 1.6. The elements Al>V>Co>Cr>Sc>Ni>Ti>Zr (in decreasing order) are present in mine-derived material in lower concentrations than in background sediments. It is evident that the elements associated with the copper-gold ore from Mount Fubilan are also found in the lowland depositional sites. Those metals which are typically enriched in soils during tropical weathering are found in higher concentrations in unpolluted lowland sediments as in material from the mine, deposited in the floodplain. The comparison of mine-derived sediments from the Fly River floodplain (Table 10)

with material from the upper Ok Tedi, which consists nearly exclusively of tailings and waste rock (Table 9), gi	ves the
following results: Calcium is found in upper Ok Tedi material 4 times and sulfur 3.8 times higher as comp	ared to
lowland mine-controlled sediments.	

Parameter	Mean	SD	Median	Percen	tiles
				25	75
Na ma/ka	11021	4150	10050	8100	15000
Na mg/kg K mg/kg	20250	4139	10930	24400	13000
K mg/kg	50550	1462	27930	24400	37200
Mg mg/kg	0/33	1403	0330	5700	//00
Ca mg/kg	80117	33/96	98450	54800	115900
Al mg/kg	61462	11766	62750	54800	65800
Fe mg/kg	47512	29854	40900	29300	54200
Mn mg/kg	700	323	775	436	965
Zn mg/kg	541	450	378	182	779
Cu mg/kg	1523	976	1158	805	1791
Pb mg/kg	463	744	123	63	488
Cd mg/kg	1.6	1.3	1.0	0.6	2.2
Au μg/kg	426	381	266	107	609
Ag mg/kg	2.4	2.5	1.4	0.8	3.3
As mg/kg	14.9	11.5	11	5	19
Cr mg/kg	30	12	26	21	33
Mo mg/kg	38	14	36	27	45
Co mg/kg	13	10	11	5	16
Ni mg/kg	17	9	14	10	23
V mg/kg	111	21	105	99	120
Ti mg/kg	1788	305	1750	1500	1900
Zr mg/kg	22	11	19	13	29
La mg/kg	23	6	22	19	26
Ba mg/kg	410	202	368	275	599
Sr mg/kg	576	86	548	510	620
Sc mg/kg	7.3	2.4	7	5	9
С%	1.8	0.5	1.9	1.3	2.1
S mg/kg	15211	20643	7900	4425	10625

Table 9. Mean, standard deviation, median values and 25% - 75% confidence intervals for 27 elements in sediments from the upper Ok Tedi, mainly tailings and waste rock (n = 24).

Factors for other important elements which are found in the Mount Fubilan orebody are: copper (2.2), silver and cadmium (both 2), zinc and strontium (1.8), gold (1.7), arsenic and molybdenum (1.6) and lead (1.5). The elements K>Ba>Mg>La>V>Ti>Ni>Sc>Zr (in decreasing order) are present in tailings and waste rock in lower concentrations than in mine-derived sediments of the Middle Fly region.

Evaluation of relationships between elements in the data obtained for the upper Ok Tedi sediments give the result that sulfur versus Fe, Mn, Zn, Cu, Pb, Cd, Ag and Mo is strongly positively correlated (r = 0.80-0.99, significance level p < 0.05). This can be explained by the fact that most metals are dicharged in sulfidic form into the Ok Tedi River. In the Fly River floodplain sediments, the positive relationship between trace metals and sulfur is much weaker (in the range of r = 0.31-0.71 for the metals mentioned above) with the exception of iron, which shows no correlation with sulfur in lowland mine sediments.

Parameter	Mean	SD	Median	Percen	tiles	
1 41 41.000				25	75	
Na mg/kg	8010	3218	8200	6025	98 00	
K mg/kg	28807	11610	30300	18850	36950	
Mg mg/kg	7242	1831	7400	6200	8575	
Ca mg/kg	0139	23842	24800	8200	50350	
Al mg/kg	0994	14915	81100	72325	91025	
Fe mg/kg	0004	11012	39500	32200	46850	
Mn mg/kg	528	254	510	310	710	
Zn mg/kg	211	68	204	163	245	
Cu mg/kg	530	289	529	278	732	
Pb mg/kg	79	36	80	48	104	
Cd mg/kg	0.5	0.3	0.5	0.2	0.7	
Au μg/kg	166	81	160	120	205	
Ag mg/kg	0.7	0.5	0.7	0.3	1.0	
As mg/kg	8.3	5.8	7.0	4.0	12	
Cr mg/kg	48	14	46	39	56	
Mo mg/kg	24	13	23	13	32	
Co mg/kg	10	5	10	7	12	
Ni mg/kg	21	9	19	15	24	
V mg/kg	145	35	139	123	163	
Ti mg/kg	2753	801	2600	2200	3200	
Zr mg/kg	44	16	41	32	52	
La mg/kg	26	7	27	22	30	
Ba mg/kg	415	100	420	363	466	
Sr mg/kg	330	139	313	226	438	
Sc mg/kg	13	4	12	10	15	
С%	1.6	1.5	0.9	0.6	2.0	
S mg/kg	2464	2160	2100	675	3425	

Table 10. Mean, standard deviation, median values and 25% - 75% confidence intervals for 27 elements in minecontrolled sediments from the Middle Fly River floodplain (n = 197).

In sediments from the upper Ok Tedi, the metals Fe, Mn, Zn, Cu, Pb, Cd, Ag and Mo are all highly intercorrelated (r = 0.65-0.95). As in the case of sulfur, the positive relationship is lost or becomes weaker in the Fly River floodplain data for the same metals. It is interesting to note that gold shows no relationship with any of the trace metals mentioned above in the upper Ok Tedi sediments. In the mine-affected lowland sediments exists a positive correlation with lead (r = 0.68) and molybdenum (r = 0.66).

There are mainly two processes operating in the river system which are responsible for the differences in the element concentrations in upper Ok Tedi and lowland sediments: Sediment admixture/erosion and mobilization of metals from the solid phase. The Ok Tedi River on its 200 km long way to the Fly River junction receives water and suspended sediments from several tributaries. Further dilution with uncontaminated riverine particulate matter occurs at the Ok Tedi/Upper Fly River confluence. Due to the fact that unpolluted tributaries have a much lower suspended load than the Ok Tedi, dilution effects, however, are small. Lateral erosion in the highly sinuous channel of the Middle Fly appears to be a more important process. Admixture of weathered floodplain sediment in the river course is responsible for increasing concentrations of elements like zirconium, scandium, titanium and chromium (which are highly persistent in chemical weathering) in the mine-derived sediments deposited downstream in the Middle Fly floodplain.



Fig. D1. Copper distribution in alluvial sediments of the Middle Fly River floodplain. The probability graph of the composite population of 385 data points (open squares) separates into two approximately lognormal subpopulations (closed squares) with a natural background population of about 50 ppm Cu, and a second population of mine-derived material at 610 ppm (geometric means).



Fig. D2. Gold distribution in alluvial sediments of the Middle Fly River floodplain. The probability graph of the composite population of 27 data (open squares) separates into two approximately lognormal subpopulations (closed squares) with a natural background population of about 7 ppb Au, and a second population of mine-derived material at 170 ppb.

A sediment particle which is discharged into the upper Ok Tedi travels about 7 days until it reaches Everill Junction 610 km downstream of the discharge point. During this time, mineral dissolution occurs. It is evident that particulate, mine-derived calcite is being dissolved in Ok Tedi and Fly River water. Particulate sulfide minerals, mainly pyrite and chalcopyrite, are also unstable in the oxygenated river waters and are oxidized to sulfates. The trace metals associated to these minerals either go in solution or become adsorbed to particulate matter, mainly to unsoluble iron oxyhydrates or organic matter. They are partitioned between the solid and dissolved phase according to their geochemical mobility in the aquatic environment of the Ok Tedi/Fly River system.

Sediment core 1/10.4.	Ca	Cu	Pb	S		
section	ppm	ppm	ppm	ppm		
0-2 cm	35400	663	51	2300		
2-4 cm	44300	668	54	1700		
18-20 cm	50000	664	98	nd		
28-30 cm	56800	618	78	2700		
30-32.4 cm	50200	574	94	nd		
32.4-34.5 cm	35200	483	83	<100		
Sediment core 5/27.3.	Ca	Cu	Pb	Au	S	
section	ppm	ppm	ppm	ppb	ppm	
19.5-22 cm	48700	977	76	205	5300	······
30-32 cm	56000	929	79	nd	3000	
46-48 cm	47800	891	71	nd	4600	

Table 11. Typical vertical profiles of mine-controlled sediments (fraction $<20 \ \mu$ m) in the Middle Fly River floodplain dTeposited at sites close to the river at high deposition rates.

Sediment core 4/7.4.	Ca	Cu	Pb	Au	S	
section	ppm	ppm	ppm	ppb	ppm	
0-2.5 cm	7800	583	92	130	600	
2.5-5 cm	6600	576	122	160	300	
7-9 cm	4200	557	140	nd	nd	
19-21.5 cm	6400	37	16	<2	<100	
Sediment core 1/7.4.	Ca	Cu	Pb	Au	S	
section	ppm	ppm	ppm	ppb	ppm	
7.5-10 cm	8700	401	102	310	200	
27-29 cm	14000	356	79	291	500	
33-35 cm	7500	69	29	nd	200	
Sediment core 2/27.3.	Ca	Cu	Pb	Au	S	
section	ppm	ppm	ppm	ppb	ppm	
3.5-5.5 cm	6100	430	110	240	<100	
5.5-8 cm	6900	320	81	nd	100	
30-32.5 cm	5900	40	19	nd	100	
Sediment core 1/30.3.	Ca	Cu	Pb	s		
section	ppm	ppm	ppm	ppm		
2-4 cm	7300	409	81	100		
26-28 cm	5800	35	14	400		

Table 12. Vertical profiles of mine-controlled sediments (fraction $<20 \ \mu$ m) deposited at swampy sites on the Middle Fly River floodplain.

Table 11 shows analytical data for two sediment cores sampled from locations close to the Fly River channel (Kai River channel, 1/10.4., and Fly River bank at Obo, 5/27.3.), whereas Table 12 shows data from swampy sites (locations in Table A3).

Calcium and sulfur values are much higher than in the cores shown in Table 12. Very little calcite dissolution and sulfide oxidation has taken place in the material from the river channel. The declining copper values towards the base of the sediment core 1/10.4. reflect the development stages of the mine. In the early mining phase of gold extraction only, tailings and waste rock consisted mainly of oxidized gossan material with a relatively low copper content and almost no sulfide minerals, which can also be seen in the base section of core 1/10.4. It was frequently observed that copper values increased in steps from the base to the top of a sediment core whereas gold showed the opposite behaviour (Table A3 in annex).

Sediments of Table 12 display very low calcium and sulfur values, although high lead and gold contents indicate that it is largely mine-derived material. Gold is a very resistant element in tropical weathering, and lead also shows very little mobility in the Ok Tedi/Fly River environment (see following section). Copper, together with calcium and sulfur, is clearly depleted in the sediments in Table 12.

Ca and S remain in the sediment body when the deposition rate is high and each layer of riverine suspended matter is rapidly covered by fresh sediment. Sulfide minerals are stable under conditions of oxygen deficiency, which are generated by the decay of riverine organic matter. The pore water in the fine-grained sediment will soon be saturated with calcium which prevents further calcite dissolution. Mobilization of trace metals will be minimal under these conditions.

To the contrary, mine-derived material deposited sporadically on swampy floodplain sites is subject to intense leaching. The undulating water table permits penetration of atmospheric oxygen into the sediment body, which is facilitated by the roots of floodplain vegetation. Both factors result in sulfide oxidation. Rainwater leaches calcite from the deposited material, and the rotting of swamp vegetation generates acidic pore waters which may bring trace metals in solution.

6.2. Hydrochemistry of the Ok Tedi/Fly River System

6.2.1. General

Earth alkaline and alkaline metals. These elements have a high solubility of their salts in common, which is independent of redox changes and partly independent of pH (Na, K). Because of the reaction:

 $CaCO_3 + H_2CO_3 \rightleftharpoons Ca^{2+} + 2 HCO_3^{-}$

calcite dissolution requires the presence of carbon dioxide, which forms the carbonic acid consumed in the reaction. The carbon dioxide partial pressure in water decreases with increasing temperatures. Dissolved Ca, Mg and Sr compounds are stable in water in the presence of carbon dioxide. Calcite is abundant in the mine waste discharged into the Ok Tedi/Fly River. Its dissolution exerts a buffering effect on pH which will remain in the range of about 6.4 to 8.0, in spite of concomitant dissolution of sulfides from mine discharge. Calcium forms complexes with humic acids. Mg compounds are, in general, more soluble than their Ca counterparts. Na is the most mobile metal in the aquatic environment. K is usually present in lower concentrations than Na. The natural leaching of K from minerals does not result in significantly increased solubilization because K is usually readily incorporated into new mineral structures like clays.

Iron and manganese. Both metals show a similar hydrochemical behaviour. Manganese tends to be more mobile as compared to iron.

Compounds of ferric iron (Fe³⁺) and Mn⁴⁺, which are the stable oxidation states in aerobic waters, are nearly insoluble. In reducing waters, the divalent reduced forms (ferrous/manganous) can persist in the absence of sulfide and carbonate anions. Iron and manganese may also occur in both oxidation states in inorganic or organic complexes. In oxygenated, alkaline waters both metals are present as colloidal suspensions of ferric resp. manganic hydroxide particles which pass through a 0.45 μ m membrane filter. For this reason, the "dissolved" concentrations reported for iron and manganese in neutral or alkaline waters in most cases do not correspond to electrolyte solutions.

Trace metals. Copper, zinc, molybdenum, cadmium and lead were analysed in waters.
Between pH 6.5 and 8.5, **lead** speciation in the aquatic environment is dominated by carbonates and hydroxides (Hem and Durum, 1973), which are almost insoluble. Pb may be complexed with organic ligands, yielding soluble, colloidal and particulate compounds. It can easily be adsorbed to particulate organic and inorganic matter, which is the dominant mechanism controlling the distribution of lead in the aquatic environment above pH 6 (Farrah and Pickering 1977).

Copper generally shows a higher geochemical mobility than lead. The distribution of copper species in the aquatic environment depends on pH and on the presence of inorganic and organic ligands. Carbonates and hydroxy-anions are the favoured inorganic complexing ligands in oxidized freshwaters with pH above 7. In the presence of soluble organic matter, sorption of copper to particulates may be relatively ineffective because complexation with humic acids is the dominant process (Jackson and Skippen 1978).

Cadmium and **zinc** are characterized by a lower affinity to carbonate and hydroxy anions at near neutral pH values. The influence of pH and alkalinity on the solubility of Cd^{2+} and Zn^{2+} is much lower as compared to lead and copper. Zinc also forms complexes with humic substances, a process which is favoured by increasing pH.

Molybdenum occurs in oxidized waters as molybdate (MoO_4^{2-}) and bimolybdate ($HMoO_4^{-}$) anions. The anions are readily adsorbed by iron and aluminium oxyhydroxides at pH values below 5. Above this value, molybdenum in natural waters is essentially dissolved.

Aluminium in the aquatic environment is mainly present in the form of dissolved and colloidal aluminium hydroxide, $Al(OH)_3$. Its minimum solubility is at pH 5.5-6. Above pH 6.5, soluble aluminium exists primarily as $Al(OH)_4$. It is capable of forming complex ions with inorganic and organic substances.

pH. The pH value in the Ok Tedi/Fly River system is not only controlled by the dissolved carbon dioxide concentration in waters and calcite dissolution, but also by the biochemical processes of photosynthesis and respiration. Photosynthesis of subaquatic plants is accompanied by the assimilation of ions such as NO₃⁻, NH₄⁺ and HPO₄²⁻. Charge balance is maintained by the uptake or release of H⁺ or OH⁻, which lead to alkalinity changes.

Alkalinity increases in oxygenated waters as a result of photosynthetic nitrate assimilation according to the bulk reaction (Stumm and Morgan 1981):

106 CO₂ + 16 NO₃⁻ + HPO₄²⁻ + 122 H₂O + 18H⁺ \ge C₁₀₆H₂₆₃O₁₁₀N₁₆P₁ (mean algae composition) + 138 O₂ NH₄⁺ is the dominant nitrogen species in reducing waters. NH₄⁺ assimilation during photosynthesis causes a decrease in pH:

 $106 \text{ CO}_2 + 16 \text{ NH}_4^+ + \text{HPO}_4^{2-} + 108 \text{ H}_2\text{O} \ge \text{C}_{106}\text{H}_{263}\text{O}_{110}\text{N}_{16}\text{P}_1 + 107 \text{ O}_2 + 14 \text{ H}^+$

Respiration of plant material leads to reactions in opposite direction. When photosynthesis and respiration are in overall equilibrium, no change in pH will be observed, although there exists a pH variability over the day/night cycle. When the rate of production of organic matter (assimilation of NH_4^+) is larger than the rate of decomposition, alkalinity will decrease. Peat formation, which was frequently observed in the Fly River floodplain, leads to low pH values in the overlying water.

Reduced species. NH_4^+ , HS⁻ and NO_2^- were determined in waters as indicators of reducing conditions. All three compounds are unstable in oxygenated water. Nitrite is metastable in both reducing (conversion to N₂ and N₂O) and oxidizing waters (nitrate formation). The stability of NH_4^+ and HS⁻ is pH-dependent. HS⁻ is converted to volatile H₂S at pH values below 7. To the contrary, NH_4^+ forms volatile NH_3 at pH above 9 (at 30°C). Within the pH range observed in the investigated waters with oxygen deficiency, only NH_4^+ was a reliable indicator of reducing conditions. Many redox reactions are slow and depend on biological mediation, hence the concentrations of species encountered in natural waters may be far from those predicted thermodynamically (Stumm and Morgan 1981).

Only the elements C, N, O, S, Fe and Mn are important participants in redox processes in the aquatic environment. However, since the mobility of trace metals is influenced by the adsorption to Fe and Mn oxides and fixation in reduced species (e.g. formation of metal sulfides), redox conditions are of importance to predict trace metal behaviour.

Dissolved organic carbon (DOC). The term DOC is a sum parameter for a number of polymeric organic substances which contain a sufficient number of hydrophile functional groups (-COO, -NH₂, R₂NH, -RS-, ROH, RO-) to remain in solution despite their molecular size. Polypeptides, amino acids, certain lipids, polysaccharides, humic and fulvic acids and Gelbstoffe belong to this group. Humic substances in general are a result of the transformation of biogenic material. DOC levels in interstitial waters of deposited organic debris may be much higher than in the overlying water.

Humic acid is extracted from humic matter in alkaline solution. Fulvic acid is the humic fraction that remains in acidified solution and is soluble over the entire pH range. Humin is the fraction which cannot be extracted by either acids or bases. Structurally, the three groups are similar; differences are in molecular weight and functional group content. The analytical determination of soluble chelates in natural waters is very difficult, particularly with the minute quantities of metal ions that are usually present.

Humic substances have a strong tendency to become adsorbed on inorganic surfaces like hydrous oxides and clays through a mechanism involving ligand exchange of humic anionic groups with H_2O and OH^- of mineral surfaces (Tipping 1981). These negatively charged coatings may themselves become active absorbers of trace metals. Colloidal iron and aluminium oxides are stabilized by humates. In highly dispersed form, these colloids pass through an 0.45 μ m membrane filter.

The most important feature of humic and fulvic acids is there tendency to form complexes with metal ions. As polycarboxylic acids, humates precipitate in the presence of dissolved Ca and Mg due to coagulation. Humic substances display colloid-chemical behaviour.

In the present study, three types of water were distinguished for their different chemistry and main constituents: Fly River water, unpolluted floodplain waters, and mixed waters.

6.2.2. Fly River Water

Water from the Middle Fly River (Table 13 and Table A4 in annex) is characterized by a moderately high content of earth alkaline and alkaline metals which is due to the active mineral dissolution of freshly eroded rock material which is carried in suspension from the Mount Fubilan minesite and the Ok Tedi catchment.

The major anion is bicarbonate, followed by the minor anions sulfate, chloride and nitrate. The dominating dissolved electrolytes are calcium and bicarbonate, which account for approximately 90% of conductivity (calculated from mol equivalents), and which also control the moderately alkaline pH of 7.7. The content of dissolved organic carbon (DOC) is fairly high at about 6 mg/l (low number of samples, not reported in Tables). Oxygen saturation measurements of Fly River water gave a mean value of only 66% which is obviously influenced by the oxygen-consuming decay of dissolved and particulate riverine organic matter. Reduced species (NH₄⁺, HS⁻, NO₂⁻) were present at or below detection limit which indicates oxygenated conditions.

Within the temperature range observed in Fly River water, p_{CO2} does not change to an extent which would markedly affect calcite dissolution. The plot of pH versus calcium (Fig. 1) shows a significantly negative correlation (p <0.05). The plot of suspended solids versus calcium (Fig. 2) displays a significantly positive relationship. It can be concluded that at low pH, more calcite from the suspended particulate fraction is dissolved, which means that the dissolved calcium concentration is controlled by pH (assuming constant atmospheric p_{CO2}).

The fact that the highest suspended solid concentrations are associated with the lowest pH values may indicate that high flows in the Fly River are associated with low pH. Since high flows are a result of heavy rainfall, and because rainwater is saturated with atmospheric CO_2 , the increased input of carbonic acid may explain near neutral pH values in the river water at high flows.

There exists a positive relationship between conductivity and suspended solids (p < 0.05, Fig. 3). A high concentration of suspended matter is associated with high values for earth alkaline and alkaline metals (Ca, Mg, Sr, Na and K).

No correlation was found for suspended solids versus-Fe, Mn and Mo.-A negative correlation between suspended matter and dissolved copper and zinc was observed, however statistically not significant. Na, Ca, Mg and Sr are highly intercorrelated.

Sample W1/21.2., which has the highest Fe value of $339 \mu g/l$, also gave the highest values for dissolved Zn and Cu. In the acidified water sample, the colloidal iron oxide particles are dissolved and adsorbed trace metals are released. This explains the high Zn and Cu values associated with iron. This correlation was not observed in other samples. Dissolved trace metal levels, with the exception of copper (median value: $17 \mu g/l$) are generally low due to alkaline pH and high bicarbonate content. Solubility of inorganic copper species in the pH range measured in the river water is very low. The fact that, in spite of the high concentration of suspended matter (which offers adsorption sites), dissolved copper values are relatively high, points to the presence of soluble organic copper complexes.



Scatter plots for selected parameters in Fly River water (Fig. 1-3) and floodplain waters (Fig. 4).

Analytical data from the pre-mining period (Maunsell 1982, Kyle 1988) are of poor quality and make comparison with present data difficult. Dissolved calcium concentrations are reported to have been in the range of 13 to 16 mg/l, which corresponds to about half of the present values. Fly River water has a high natural content of earth alkaline and alkaline metals because of the limestone formations mainly in the Ok Tedi catchment. Fly River water chemistry probably was not much different from present day conditions, although Maunsell (1982) and Kyle (1988) report slightly acidic pH values in the range of 5.5 to 6.7 which appear erroneous. Trace metal levels measured by Maunsell (1982) were at or below the detection limit of 1 μ g/l with the exception of Fe, Mn and Zn. Pickup et al. (1979) report mean suspended solid concentrations in the range of 60 to 80 mg/l for the Middle Fly and note that "the Fly and the lower Ok Tedi are very clean rivers by Papua New Guinea standards".

Parameter	Mean	SD	Median	Percen	tiles	
				25	75	
T ()0	28.2	17	28.0	26.4	20.4	
Temperature °C	28.2	1.0	28.9	26.4	29.4	
pH	7.7	0.2	7.7	7.7	7.8	
Conductivity µS/cm	138	15	136	128	146	
Oxygen Saturation %	66	0.2	64	60	68	
Suspended Solids mg/l	199	149	155	92	207	
Na μg/l	1564	299	1425	1350	1660	
K μg/l	622	87	615	536	667	
Ca μg/l	29380	6654	26000	24750	31300	
Mg μg/l	1237	170	1190	1075	1345	
Sr µg/l	172	36	158	144	181	
Al μg/l	below de	etection lin	nit of 50 µg/l			
Fe μg/l	72	102	30	8	75	
Mn μg/l	18.4	13.2	12.5	8.5	25	
Zn μg/l	9.2	6.2	8.0	4.0	10.5	
Cd µg/l	below de	etection lin	nit of 0.1 µg/	I		
Cu µg/l	19.6	11.8	17.0	13.0	19.3	
Pb μg/l	below de	etection lin	nit of 1 µg/l			
Mo μg/l	7.9	5.0	7.0	3.0	12.0	
HCO ₃ mg/l	7 7	4.0	76	73	78	
SO ₄ mg/l	5.0	2.5	3.5	0.0	6.0	

Table 13. Mean, standard deviation, median values and 25% - 75% confidence intervals for Fly River water samples (n = 11).

6.2.3. Floodplain Waters

The term floodplain waters is used for courses of water which drain the outer Fly River floodplain. Their catchment comprises forested areas and low-lying swamps of high biological productivity with dense subaquatic and swamp vegetation. The main features of floodplain waters are low pH and conductivity (Table 14 and Table A5 in annex) and their yellow-brown colour ("blackwater").

The content of suspended solids is very low as compared to the Fly River. The material retained on the membrane filter is orange brown, consisting of iron oxides (about 10-15% Fe) and particulate organic matter. The high content of iron, most probably in the form of oxyhydrate colloids, is particularly evident in the non-filtered water samples, but also in the filtered water in which the iron values are much higher than in the Fly River water. Sodium, an ubiquitous element, is present in floodplain waters in a similar concentration range as in the Fly River. Aluminium values were highest at low pH and conductivity (Fig. 4). The overall Al content was higher than in the Fly River, where the metal was below the detection limit (50 μ g/l) in the four samples measured.

An influence from the mine discharges was detectable in some water samples (slightly elevated trace metal, calcium and sulfate levels). However, since the main features of floodplain drainage such as low pH and conductivity and high DOC levels were observed in these samples, they were included in this category. Metal levels for Cu, Pb, Cd and Mo in floodplain waters unaffected by mining are at or below detection limit (e.g. $<2 \mu g/l$ for Cu, Table 14). Zinc is the only heavy metal present in measurable concentrations. Since floodplain waters drain lowland areas of intensely weathered Pleistocene sediments, extremely low trace metal are to be expected.

Low alkalinity is a result of active accumulation of organic material. Peaty sediments were repeatedly encountered during lake bottom coring. Oxygen deficiency in stagnant or slowly moving, warm waters does not allow complete respiration (oxidation) of organic matter. There is a positive correlation between temperature and oxygen saturation (Fig. 5), which is contradictory at first sight. However, the oxygen measured was a result of photosynthetic activity, which is highest at intense sunshine, which in turn is responsible for high water temperatures. It is obvious that the waters are not in redox equilibrium. The photosynthetic oxygen production masks the overall oxygen deficiency of the system, which is evident from the relatively high concentrations of reduced species like NH_4^+ and HS^- . Of both compounds, NH_4^+ is the better indicator of reducing conditions because hydrogen sulfide is volatile at the low pH and high water temperatures measured. NH_4^+ shows a significantly positive correlation with dissolved organic carbon (DOC)(Fig. 6), which is also positively linked with water temperature. NH_4^+ and DOC are both a result of decomposition of organic matter. The DOC produced in off-river sites is the main source of organic ligands which are responsible for trace metal complexation in polluted waters.

Parameter	Mean	SD	Median	Percenti	es
				25	75
Temperature °C	30.6	2.2	30.0	29.2	31.4
рН	6.0	0.5	6.0	5.7	6.4
Conductivity µS/cm	28.5	10	27	20	34
Suspended Solids mg/l	21	11	19	11	24
Na µg/l	1233	409	1140	1020	1280
K μg/l	188	107	193	89	244
Ca µg/l	4295	1992	4110	3060	5890
Mg μg/l	477	141	480	372	499
Sr µg/l	27	10	24	21	34
Al µg/l	67	48	64	25	80
Fe μg/1	values hig	hly variable	:		
Mn μg/l	15.4	15.1	7.5	2.5	27
Zn µg/l	close to de	etection limi	it of 5 µg/l		
Cd µg/l	below dete	ection limit	of 0.1 µg/l		
Cu µg/l	close to de	etection limi	t of 2 µg/l		
Pb µg/l	below dete	ection limit	of 1 µg/l		
Mo μg/l	below dete	ection limit	of 5 µg/l		
HCO ₃ mg/l	13	6.5	13	9	17
NH ₄ mg/l	0.3	0.2	0.20	0.14	0.38
SO ₄ mg/l	0.37	0.26	0.27	0.18	0.45
DOC mg/l	9.4	3.0	8.0	7.0	11.0

Table 14. Mean, standard deviation, mean values and 25% - 75% confidence intervals for outer floodplain water samples (n = 15).

6.2.4. Mixed Waters

Mixed waters are generally intermediate in composition between Fly River and floodplain waters, although the influence of the Fly River is clearly dominant (Table 15 and Table A6 in annex).

Due to the generally flat terrain, the location of the mixing zone of floodplain drainage and Fly River water depends mainly on rainfall in the upper catchment of the Fly River/Ok Tedi and in the Fly River lowland. When there is high rainfall in both catchment areas, the mixing front between Fly water rich in suspended solids and blackwater from the floodplain will be located close to the mouths of creeks and tie channels.



Scatter plots for selected parameters in floodplain waters (Fig. 5-6) and mixed waters (Fig. 7-8).

At high flow conditions in the Fly River and previously little rainfall in the lowland, river water intrudes several kilometers upstream of the channels and lakes which drain into the main river under reverse conditions. The intrusion of Fly River water is associated with transport of mainly mine-derived suspended matter, which is deposited in the waters of the inner floodplain. Hence, it is not possible to distinguish between dissolved metals in mixed waters which are directly derived from an intrusion of Fly River water and those that may be mobilized secondarily from the sediments already deposited.

Because of the different composition of Fly River waters and those draining the outer floodplain, the physical and chemical interactions occuring are of environmental interest (Fig. P8). Particular attention has to be paid to the behaviour of trace metals, of which copper is the most relevant.

Parameter	Mean	SD	Median	Percen	tiles	
				25	75	
Temperature °C	30.2	2.0	30.1	28.9	31.7	
pH	7.3	0.8	7.1	6.6	7.7	
Conductivity µS/cm	130	90	119	80	151	
Oxygen Saturation %	7 7	42	77	41	107	
Suspended Solids mg/l	36	72	15	7	28	
Na μg/l	15 8 7	541	1480	1185	1755	
К µg/l	539	657	385	272	603	
Ca µg/l	21390	8372	21900	13425	28050	
Mg µg/l	1226	725	1140	881	1345	
Sr µg/l	150	111	138	82	176	
Al µg/l	102	139	25	25	101	
Fe µg/l	357	590	145	32	457	
Mn µg/l	82	308	2.5	2.5	24.5	
Zn ug/l	15.4	15.5	12	6	19	
Cd µg/l	close to	detection l	imit of 0.1 μ	g/l		
Cu µg/l	11.8	11.3	9	5	14	
Pb ug/l	close to	detection l	imit of 1 µg/	1		
Mo ug/l	7.8	9.9	5.0	2.5	9.0	
HCO ₂ mg/l	53	28	47	39	62	
NH₄ mg/l	0.21	0.16	0.15	0.10	0.25	
SO ₄ mg/l	2.5	2.0	2.1	0.7	3.4	
DOC mg/l	9.0	2.5	9,0	7.3	11.0	

Table 15. Mean, standard deviation, median values and 25% - 75% confidence intervals for mixed water samples (n = 65).

As mentioned above, the pH of floodplain waters is much lower than in the Fly River. Decreasing alkalinity may lead to increased heavy metal mobility due to desorption from particulate matter and formation of free dissolved metal species.

The plot of pH versus Ca shows a weakly positive correlation (Fig. 7). Mg, Sr, Na, K and HCO₃⁻ give similar plots. Calcite in particulate form carried into the floodplain waters is rapidly dissolved and exerts a buffering effect on the local waters. Opposite to the main trend in the pH/Ca plot, there are samples showing a high calcium content at comparatively low pH. These data are from small ponds on the swampy floodplain and slightly acidic floodplain seepage, i.e. extremely iron-rich water trickling from exposed channel or river banks into the river during low flow conditions.

No clear influence of pH on dissolved iron and manganese concentrations was observed, although the highest Fe values tend to be associated with low pH.

None of the trace metals showed a significant correlation with pH. The plots of pH versus zinc (Fig. 8) and cadmium display a slightly negative trend. No correlation between pH and Cu (Fig. 9) and pH and Mo was detected. DOC and pH are weakly negatively correlated (Fig. 10) which is due to the acidic nature of humic and fulvic substances. In the absence of calcium bicarbonate buffering, the organic acids control water pH.



Scatter plots for selected parameters in mixed waters (Fig. 9-12).

Reducing conditions, indicated by an elevated NH_4^+ content, also seem to have little influence on dissolved trace metal concentrations. Only zinc gave a moderately positive correlation with NH_4^+ .

Ca shows a weakly positive correlation with Mo (Fig. 11). A positive relationship of the alkaline and earth alkaline metals with sulfate (calcium versus sulfate, Fig. 12) was observed. Elevated concentrations of Ca, Mg, Na, K, and Sr are clear indicators of the influence of mine discharges. High sulfate levels are a product of active dissolution of minederived sulfide minerals. Although the metals iron, zinc, copper and molybdenum are discharged into the river system primarily in the form of sulfides and undergo oxidation to sulfates, no positive correlation between sulfate and any of the metals was observed. Iron and zinc even displayed a weakly negative relationship with sulfate. This illustrates the complexity of solution chemistry in the investigated waters, and the fact that aqueous metal transport is not controlled by sulfate complexation. There was also no significant correlation of dissolved organic carbon (DOC) and trace metals found. Only the plot of zinc versus DOC shows a weakly positive correlation.



Fig. P8. Mixing of acidic, humic-rich floodplain drainage (blackwater, DOC 12 mg/l, 18 μ g/l Cu) and sediment-laden white water of the Fly River (DOC 5 mg/l, 9 μ g/l Cu) in the Lake Bosset channel.

Intercorrelations between elements in mixed waters are very high in the group Ca, Mg, Sr, Na, K and HCO₃⁻, with the exception of Na versus K. Zn and Cd display a positive correlation, too. A significantly positive correlation was found for Al and Cu in unfiltered water samples (Fig. 13). Aluminium hydroxide may complex or adsorb dissolved copper. Al/Zn displayed a similar correlation. Fe versus Cu in unfiltered water shows no clear trend. Fe is weakly positively correlated with DOC and Mn.

The heavy metal concentrations in waters of the Fly River inner floodplain, the zone of mixed water, are of particular interest because of the prominent role which off-river waters play in the ecology and biological productivity of the entire river system.

Dissolved trace metal levels in mixed waters are controlled by a number of abiotic and biotic factors. The most important inorganic factor is the moderately high bicarbonate content of Fly River water and the high earth alkaline metal concentrations, dominated by calcium in dissolved and particulate form, which are responsible for the neutral to alkaline pH values in mixed waters. The most prominent biotic factor are the dissolved organic carbon substances which play a very important role as complexing agents.

Both factors interact in a complicated manner which cannot be predicted from thermodynamic equilibrium calculations.

Despite the fact that lead in mine wastes is clearly enriched above background values, the dissolved metal contents were in the great majority of samples below detection limit (< $1 \mu g/l$). Elevated concentrations were found only in unfiltered samples and in a few samples which also showed a high content of presumably colloidal iron and manganese, which offer adsorption sites for lead.

Speciation modelling of inorganic lead with PHREEQE suggests that at near neutral pH values and oxic conditions in waters, more than 70% of soluble lead exists in the form of the $PbCO_3^0$ complex which easily precipitates out. Between 0-30% of lead may be present in the free ionic form Pb^{2+} under the environmental conditions in the floodplain.



Scatter plot for Al versus Cu in mixed waters (Fig. 13).

Cadmium and zinc are among the geochemically most mobile elements. Their mobility is less pH dependent as compared to lead. Cd was found in concentrations above detection limit (> $0.1 \mu g/l$) only in waters with a pH below 7. Because of the low Cd values in mining residues, the metal is not considered as being of environmental concern. The same holds true for zinc, which showed a similar concentration range in all waters investigated.

Molybdenum, because of its anionic form in water, showed a different behaviour than the other trace metals. Highest concentrations were usually found in alkaline waters, or linked with high calcium values. Mo as well as Zn is considered an essential element to biota. The concentrations observed are much below the toxic threshold.

Copper is an essential element to plants and animals, too.

Because of its environmental importance, the fate of copper in the investigated waters will be discussed in some detail in the following chapter.

6.2.5. The Behaviour of Copper

The statistical data analysis showed that copper concentrations in water do not appear to be significantly controlled by inorganic factors.

Alkalinity and pH, adsorption to oxide surfaces (with the exception of aluminium) and reducing conditions (e.g. formation of soluble copper amine complexes) do not seem to have a significant effect on dissolved copper levels. The abundance of organic complexing agents in the investigated Middle Fly floodplain waters may be responsible for the observed behaviour of dissolved copper. No correlation was observed between DOC and copper levels in the floodplain waters. It is evident that dissolved organic ligands, even at comparatively low concentrations in waters of the Fly River system, are always present in excess of trace metals which may become complexed (although some complexing capacity may be occupied by major cations like Ca and Mg). In the Lower Fly River, downstream of the Strickland junction, measured dissolved copper values are still moderately high at 6 µg/l (Table A6, annex) despite the massive admixture of uncontaminated suspended matter from the Strickland River. It appears that the lowering of dissolved copper levels in the Lower Fly is mainly due to dilution effects and that adsorption to riverine particulate matter is of secondary importance.

Davis and Leckie (1978) tested the influence of dissolved organic substances on copper adsorption. Certain organic ligands which form coatings on suspended mineral particles enhance the extent of trace metal adsorption, while others show opposite behaviour. Some organic acids inhibit copper adsorption by forming soluble, stable complexes which keep the metal in solution. When the complexing capacity of dissolved organic carbon substances is larger than the rate of adsorption to inorganic surfaces, copper will remain in solution.

Sholkovitz and Copland (1981) also investigated the solubility and adsorption properties of a number of trace metals in the presence of humic acids. Contrary to similar studies which were performed with synthetic laboratory waters, Sholkovitz and Copland used natural water from a small stream in Scotland which drains peaty soils. The properties of this water (DOC 7 mg/l, pH 6.5, low conductivity) are similar to those of the Fly River floodplain. The authors detected a coagulating effect on humic acids and trace metals, particularly iron and copper, upon adding of only 0.5 mmol/l of Ca, which is equivalent to 20 mg/l Ca. The mean Ca concentration in mixed waters of the Fly River inner floodplain is about 25 mg/l. Coagulation of humic substances with adsorbed trace metals may be an important process when calcium-rich Fly River water mixes with floodplain blackwaters. This is consistent with the detection of elevated copper levels on the top of black, peat-like sediments sampled from lake bottoms far from direct Fly River influence. The experiments carried out by Sholkovitz and Copland (1981) also gave the result that there was no precipitation of trace metals and humic acids when pH (starting point pH 6.5) was changed in the range of 9.5 to 3, below which humic acids, Fe, Mn, Cu, Ni and Cd began to precipitate. This behaviour is contrary to that predicted by inorganic solubility considerations. Complexation of the trace metals by dissolved organic matter is the most reasonable explanation. Organic copper complexes are known to be particularly stable because of their favourable electron configuration (Stumm and Morgan 1981).

Fe and Mn in floodplain waters, despite their chemical similarity, showed no significant correlation in any data set. Of the elements investigated by Sholkovitz and Copland (1981), Fe>Cu>Ni>Cd (in decreasing order) showed the strongest affinity for the dissolved humic substances, Mn and Co the least. The differing tendency of Fe and Mn to form complexes with dissolved organic carbon may explain the missing correlation in data from the Middle Fly River region. In a copper adsorption experiment (spiking to yield 20 μ g/l Cu) with natural waters containing different concentrations of suspended solids and dissolved organic carbon, Sholkovitz and Copland (1981) obtained results which suggest that solubilization of Cu by dissolved organic ligands, forming ultrafine colloids (< 0.01 μ m), is a more important process than the adsorption onto riverine particulate matter. Even in water containing 100 mg/l mostly inorganic suspended matter and 3 mg/l DOC, most of the copper was kept in solution as the pH increased from 4 to 9.

CSIRO of Australia (1989) performed mixing experiments commissioned by Ok Tedi Mining Ltd. with water from the Fly River and a floodplain tributary. Mine derived sediment with high copper values was added to different admixtures. Total dissolved copper concentrations increased with sediment admixture, the pH remained fairly constant. Dissolved copper species in the resulting solutions were analyzed by Anodic Stripping Voltametry (ASV). The method is used to determine labile trace metal species supposed to be present in the ionic, most bioavailable form. Interestingly, between 30-50% of dissolved copper in the final test solutions was measured as labile or "ionic" copper. Assuming that the copper was present as dissolved organic species, the results point to great reactivity of humic copper complexes in the Fly River system. This is consistent with more recent laboratory work undertaken by CSIRO on behalf of OTML (OTML 1994). Electrochemical measurements of bioavailable copper in the OK Tedi/Fly River system gave the result that the fraction of potentially bioavailable copper is up to 50% of the total dissolved copper concentration.

In the presumably unpolluted waters of the outer floodplain (it is difficult to establish the maximum intrusion range of Fly River water) copper was at or below the detection limit of 2 μ g/l. In the Fly River, copper values were about tenfold above this "background", which may actually be much lower than 2 μ g/l, i.e. 0.2 μ g/l.

Mixing of both waters should result in dilution. This is the case in most mixed water samples, however in some waters copper concentrations were much higher. Sample W1/7.4., which had a dissolved copper concentration of 52 μ g/l in the filtered and 106 μ g/l in the unfiltered sample, was taken from a depression in flat swampy terrain about 150 m behind the low dam paralleling the Fly River channel.

YEAR	KUA	MBI	T/NUKUM	1BA	(OBO		C)GWA
	рCud	lCu í	fish catch	pCu o	iCu	fish catch	pCu	dCu	fish catch
	ug/g	ug/l	kg	ug/g	ug/l	kg	ug/g	ug/l	kg
1990	1236	11	15	1202	17	35	524	3	95
1991	9 05	6	28	879	9	66	401	2	118
1992	715	3	38	-693	5	94	321	11	35
1993	875	4	29	850	6	70	388-	1	120
1994	581	3	49	564	4	119	270	1	147
1995	581	3	49	564	4	119	270	1	147
1996	562	3	49	544	4	119	253	1	147
to 2008									

Table 2. Particulate and dissolved copper and fish catch for Kuambit/Nukumba, Obo and Ogwa.

Source: pCu	-Supplementary Investigations, Vol. II, Appendix B.
dCu	-Supplementary Investigations, Vol. I, Figure 3.
Fish Catch	-Supplementary Investigations, Vol. III, Appendix H,
	Figures 10a, 11a, & 12a and Table 4A.

Table 18	Dissolved (lopper f	Predictions	(ug/L)	Provided	to the	State in	June	1992.
	DISSOLACT C	, upper r	. i calcuons	(222)	t romaca	10 110	0.000		

Year	Nukumba	Оро	Ogwa	
1991	11 (10.5)	15 (14.2)	6 (5.4)	
1992	13 (13.1)	17 (15.0)	6 (6.6)	
1993	14	17	7	
1994	11	15	6	
1995	11	15	6	
1996	12	18	7	
Error Estimate	+- 78%	+- 80%	+- 85%	

() annual (calendar year) means of observed data

Table 5.8 Revised Predictions for dCu in the Fly River and Mean Data for 1993

Year	Nukumba	Obo	Ogwa
1993	[16 {4 - 28}(13.8)	17 (3 - 31) (17.6)	6.3 {1 - 12} (7.4)
1994	18 [4 - 32]	18 (4 - 32)	7.5 (1 - 14)
1995	14 [3 - 25]	14 (3 - 25)	6.7 (1 - 12)
1996	14 (3 - 25)	14 (3 - 25)	6.6 {1-12}
1997	13 (3 - 25)	13 {3 - 25}	6.1 (1 - 11)
1998	14 [3 - 25]	14 {3 - 25}	6.6 (1 - 12)

() Upper and lower error estimates

() Annual (chiendar year) means of observed data

Table 16. Predictions made by OTML in 1989 for key environmental parameters in the Fly River system, and revised predictions for dissolved copper in 1992 and 1993 (OTML 1990, 1993, 1994).

Water sampled from small pools and shallow water courses in the periodically flooded grassland sites usually gave high copper values. The water samples taken generally had low pH and/or alkalinity. It appears that active leaching of copper from deposited mine-derived material is responsible for the high dissolved values observed. In the floodplain swamps, redox conditions change easily depending on the undulating water table which may result in trace metal mobilization. Because of the dense vegetation and the low rate of water exchange, soluble organic chelates are abundant and may facilitate copper mobilization. The buffering effect of calcium disappears as the easily soluble element is leached from the sediments. Because of the favourable conditions for trace metal leaching, even a thin layer of deposited mine-derived material will be an important source of copper. Due to difficult access, only a few samples from the swamp sites were taken (water was sampled within a maximum distance of one kilometer from the main river channel). It seems that the copper levels in floodplain waters investigated in the present study are biased towards low values (in samples from large lakes and tie channels) and are not representative for the entire floodplain.

Where floodplain swamp waters with leached copper drain into lakes and the main river, they will increase copper concentrations. This may explain why OTML (1991, 1993, 1994) reports higher dissolved copper levels at Obo as compared to the upstream site at Nukumba, below the Ok Tedi/Upper Fly River confluence.

A different type of water in which high trace metal concentrations could be expected are the samples called "floodplain seepage". These moderately acidic waters percolate through the floodplain sediments and drain spring-like into the channels and rivers. The waters are highly reducing and have a very high Fe^{2+} and Mn^{2+} content which is immediately oxidized to orange coloured gels and flocs upon exposure to atmospheric oxygen. Although the floodplain seepage samples showed high Cd and Pb levels, this was not the case for copper, which probably remains in sulfidic binding in the sediment body.

Table 16 shows predictions based on computer modelling by OTML (1990) of various environmental parameters in the Fly River system in response to the impact of mine discharges. These additional environmental monitoring conditions were established to ensure that the Acceptable Particluate Level of 940 mg/l at Nukumba does not result in actual environmental damage to the Fly River System beyond the level actually specified in the predictions (OTML 1990).

The predictions resulting from the Supplementary Environmental Investigations (undertaken in 1986-89 by OTML) have been established for testing whether or not the State's conditions for environmental management of the Fly River System and off-shore are met. Due to a number of reasons, there has been a non-compliance with monitoring conditions beginning in 1990. Particularly the dissolved copper levels were much higher than expected. Measured values at the APL sites Kuambit/Nukumba, Obo and Ogwa have been in a steady increase between 1990 and 1993 (OTML 1994) contrary to the trend predicted by the models. Revised prediction values for the key environmental parameters were developed by OTML in July 1992 and again in December 1993.

7. DISCUSSION OF BIOLOGICAL IMPACTS

The discharge of ore processing residues and mining wastes has significantly changed the aquatic environment of the Ok Tedi/Fly River system. From the discussion in previous sections it can be concluded that there are three factors of particular environmental concern: the increase in particulate copper, in dissolved copper, and in suspended solids concentrations. The negative impact of each effect on the fluvial system is spatially different.

The massive increase in the suspended load of the Ok Tedi is mainly responsible for the decline in aquatic life upstream of the confluence with the Fly River, where fish populations and species diversity have been dramatically reduced (Smith 1991). The persistently high concentration of suspended sediments in the water interferes with gill respiration of fish and aquatic organisms, modifies the movement and migration of fish and prevents successful development of eggs and larvae. Heavy metals in particulate and dissolved form, and toxic process chemicals like xanthates probably play a minor role in the adverse impact on the aquatic life of the Ok Tedi River.

In the Fly River itself, a combined detrimental effect of suspended sediment, particulate and dissolved copper on the aquatic ecology is to be expected. Monitoring undertaken by OTML (Smith 1991, OTML 1993, 1994) shows that fish populations in the Middle Fly River at Kuambit, immediately downstream of D'Albertis Junction, are continuously

declining, in the range of 30-50% (M. Eagle, pers. comm.). Statistical analyses gave the result that particulate copper (pCu) was the best negative correlate of fish catches, but dissolved copper and suspended solids also have a negative effect. OTML (1994) speculates that the relationship between pCu and fish catch may either be attributed to an avoidance of pCu by fish or by a correlation of pCu with a toxic fraction of dissolved copper.

Both total fish catches and the diversity of the aquatic fauna are affected. No clear negative impact of mine discharges on the abundance of fish in the Middle Fly River downstream of D'Albertis Junction can be established from the data collected so far because of their high variability. Fish biomass monitoring in the last three years has been strongly influenced by low river levels concentrating fish into the river channel as the floodplain dried, hence leading to high catches at the river sites. Clearly, the exact determination of fish populations (including migratory species) in a dynamic fluvial ecosystem is a difficult undertaking since fish catches are influenced by a number of factors which cannot be attributed to the mine waste, like periods of droughts and commercial and subsistence fishing pressure. In addition, OTML (1994) admits that its database on fish biomass is limited with respect to baseline data and comes to the conclusion that "changes between before and after mine start-up conditions cannot be quantified for any site (in the Fly River system)".

Considering the homogeneity of the Middle Fly River system and the fact that the key water parameters influenced by the mine's operation (dissolved and particulate copper and suspended solids) show little changes in the Middle Fly River reach between Kuambit/Nukumba and Obo, negative effects on the fish populations in the lower part of the Middle Fly region are to be expected.

OTML (1993) report elevated copper levels in body tissue of the catfish *Arius berneyi* caught in Lake Pangua, a deep oxbow lake. *Arius* is a bottom feeding omnivore. Aquatic invertebrates form the most important part of its diet, but detritus/mud is also an important food item (Kare 1992). It is not clear whether the copper detected in the fish was taken up via the bottom dwelling invertebrates, or directly from the copper-rich sediment, the metal being released in the intestinal tract of the catfish. In both cases, it is evident that copper in particulate form on the lake bottom is bio-available.

In the off-river floodplain sites of the Middle Fly region, elevated concentrations of dissolved copper may negatively affect aquatic life. The floodplain plays an important role in primary productivity to support fish stocks in the river system, and for the recruitment of fish. OTML (1991) states that the floodplain supports a greater stock of fish than do oxbow and drowned valley lakes. The Fly River channel is to a large extent a fish migration route and a refuge during dry periods. Subsistence and commercial fisheries target mainly the off-river water bodies because of their high productivity. Although the vegetated floodplain sites may not be inundated over the whole year, fish invade rapidly into the alluvial plain at higher water levels because of the abundance of food sources.

Animals at lower trophic levels, such as aquatic invertebrates, and juvenile stages of fish are known to have a high sensitivity to dissolved copper. According to Moore and Ramamoorthy (1984), sensitivity to copper is inversely related to the age or size of an animal. A decrease in the abundance of macroinvertebrates, which are at the base of the food web, may lead to a decrease in the overall fish population. Copper is a strong toxicant to aquatic organisms but does not biomagnify in the food chain. End consumers like carnivorous fish show a much lower copper body burden than benthic invertebrates like burrowing mayfly larvae which feed directly from the contaminated substrate (Karbe 1988). The average copper concentration in mixed waters of the inner floodplain is around 10 μ g/l; the Fly River water has about 17 μ g/l of copper. These are concentrations which may be harmful to biota. The applicable United States Environmental Protection Agency (USEPA) "Water Quality Criteria for the Protection of Aquatic Organisms and Uses" (1986) recommends a maximum value of 6.5 μ g/l (hardness 50 mg/l as CaCO₃, unfiltered water) for the "chronic" 4-day average concentration, and 9.2 μ g/l for the "acute" 1-hour average concentration. The Canadian Water Quality Guidelines (1987) are even more stringent (Table 17).

Dissolved copper levels detected in mixed floodplain waters of the present study are in the concentration range which is reported in the literature to be harmful to sensitive freshwater aquatic organisms. Clements et al. (1989) found a significant decrease in total aquatic insect abundance after 4 days of exposure to 6 μ g/l of copper in an outdoor experimental stream. The most sensitive species, a chironomid larva (midges, Diptera), was eliminated at 13 μ g/l copper after 10 days. Moore and Winner (1989) also found out that benthic chironomid and mayfly larvae show a high sensitivity to dissolved copper.

Table 3-1. Summary - Guidelines for Freshwater Aquatic Life

Parameter

pН

Selenium Silver

Thallium Zinc³

Physical parameters Temperature

Total suspended solids

Inorganic parameters

Aluminum

Guideline

0.005 mg·L-1

0.1 mg·L -

Comments

Thermal additions should not alter thermal stratification or turnover dates. exceed maximum weekly average temperatures, and exceed maximum short-

term temperatures (see Section 3.2.3.1.1)

Background suspended solids $\leq 100.0 \text{ mg} \cdot \text{L}^{-1}$

 $pH \le 6.5$; $\{Ca^{2-}\}\le 4.0 \text{ mg} \cdot L^{-1}$; $DOC \le 2.0 \text{ mg} \cdot L^{-1}$ $pH \ge 6.5$; $\{Ca^{2-}\}\ge 4.0 \text{ mg} \cdot L^{-1}$; $DOC \ge 2.0 \text{ mg} \cdot L^{-1}$

Antimony	ID ²	
Arsenic	0.05 mg·L = 1	
Beryllium	ID	
Cadmium	0.2 µg·L-1 0.8 µg·L-1 1.3 µg·L-1 1.8 µg·L-1	Hardness $0-60$ mg·L ⁻¹ (CaCO ₃) Hardness $60-120$ mg·L ⁻¹ (CaCO ₃) Hardness $120-180$ mg·L ⁻¹ (CaCO ₃) Hardness >180 mg·L ⁻¹ (CaCO ₃)
Chlorine (total residual chlorine)	2.0 µg·L ⁻¹	Measured by amperometric or equivalent method
Chromum	0.02 mg·L ⁻¹ 2.0 μg·L ⁻¹	To protect fish To protect aquatic life, including zooplankton and phytoplankton
Copper	2 µg·L=1 2 µg·L=1 3 µg·L=1 4 µg·L=1	Hardness 0-60 $mg \cdot L^{-1}$ (CaCO ₃) Hardness 60-120 $mg \cdot L^{-1}$ (CaCO ₃) Hardness 120-180 $mg \cdot L^{-1}$ (CaCO ₃) Hardness >180 $mg \cdot L^{-1}$ (CaCO ₃)
Cyanide	5.0 µg-L=1	Free cyanide as CN
Dissolved oxygen	6.0 mg·L ⁻¹ 5.0 mg·L ⁻¹	Warm-water biota – early life stages – other life stages
	9.5 mg·L ⁻¹ 6.5 mg·L+1	Cold-water biota – early life stages – other life stages
lron	0.3 mg·L ⁻¹	
Lead	1 μg·L-1 2 μg·L-1 4 μg·L+1 7 μg·L-1	Hardness 0-60 mg·L ⁻¹ (CaCO ₃) Hardness 60-120 mg·L ⁻¹ (CaCO ₃) Hardness 120-180 mg·L ⁻¹ (CaCO ₃) Hardness >180 mg·L ⁻¹ (CaCO ₃)
Mercury	0.1 µg·L - 1	
Nickel	25 μg·L-1 65 μg·L-1 110 μg·L+1 150 μg·L-1	Hardness 0-60 mg·L ⁻¹ (CaCO ₃) Hardness 60-120 mg·L ⁻¹ (CaCO ₃) Hardness 120-180 mg·L ⁻¹ (CaCO ₃) Hardness > 180 mg·L ⁻¹ (CaCO ₃)
Nitrogen		
Ammonia (total)	2.2 mg·L=1 1.37 mg·L=1 0.06 mg·L=1	pH 6.5: temperature 10°C (see Table 3-12) pH 8.0: temperature 10°C
Nitrosamines	ID	Concentrations that summate profine weed growth should be avoided

mg-L-I increase of 10% above

6.5-9.0 1 µg·L=1

0.1 μg·L-1

0.03 mg·L = 1

increase of 10.0

ID

Background suspended solids \geq 100.0 mg·L⁻¹ background Concentrations of heavy metals reported as total metal in an unfiltered sample.

 2 ID = insufficient data to recommend a guideline.

³ Tentative guideline.

Table 17. Canadian Guidelines for the protection of freshwater aquatic life (CCREM 1987).

Burrowing mayfly larvae of the genus *Plethogenesia* (together with *Macrobrachium* freshwater prawns) were the dominant constituents of the macroinvertebrate fauna in terms of biomass in the Fly River (OTML 1987). Both are important food items for the local fishes. Toxicity of particulate copper in mine wastes on mayflys has been proven by bioassays undertaken by OTML in 1988 (OTML 1989).

Williams et al. (1991) report a 50% mortality after three days of exposure of the extremely sensitive tropical freshwater shrimp *Caridina sp.* to 4 μ g/l copper. Most studies on copper toxicity consider the free ionic species as the most bio-available form, and copper complexed with naturally derived dissolved organic carbon (DOC) as much less toxic or even non-toxic (e.g. Flemming and Trevors 1989; Meador, 1991). Recent scientific work (Winner and Owen, 1991), however, has provided further evidence that dissolved organic carbon also may enhance copper toxicity. The authors used the alga *Chlamydomonas reinhardtii* to test the bioavailability and toxcity of organically complexed copper in water from a freshwater pond.

Toxic effects of dissolved copper on alga deflagellation and population growth were observed at values above 12.2 μ g/l in natural waters with DOC concentrations varying from 5 to 14 mg/l.

They found a positive correlation between DOC and copper on alga toxicity and concluded that labile organic copper complexes may be responsible for this effect, demonstrating that a simple relationship between DOC concentration and copper toxicity cannot be established. OTML (1994) has commissioned toxicity tests with the freshwater green alga *Chlorella protothecoides*. No reduction in algal growth rate occurred at dissolved copper concentrations of 12 μ g/l, which was the highest concentration used in the tests. Since copper values measured both during the present study and by OTML in waters of the Fly River and its floodplain were much higher than 12 μ g/l, it appears justified to repeat the tests with copper concentrations of up to 50 μ g/l in order to obtain more information on the species' sensitivity.

8. CONCLUSIONS

The following conclusions can be drawn from the informations gathered in the present study:

1. Of the trace metals contained in the Ok Tedi mine waste, copper is of main environmental concern because of its strong enrichment above background (about twentyfold) and its relatively high geochemical mobility.

2. The discharge of mining residues into the Ok Tedi/Fly River system leads to a substantial deposition of copper-rich material in the Middle Fly River floodplain, although most of the mining wastes carried as suspended load finally reach the delta. Areas of standing water and vegetated parts of the floodplain play an important role in the recruitment of fish and primary productivity. Hence, this part of the fluvial ecosystem is particularly sensitive to mine-induced changes in the local environment.

3. The pollution by copper-rich material is not only a problem of quantity, but also of spatial distribution and the potential mobilization of the trace metal. Deposition of mine-derived sediments in the Fly River floodplain is highest in oxbow lakes. Because of the naturally high organic carbon content in sediments, which is responsible for reducing conditions, only copper in the uppermost layer (few centimeters) of bottom sediments may become dissolved. When copper-rich sediment is permanently covered by water, mobilization and bioavailability of the metal, except to bottomdwelling invertebrates, is low. However, when mine-derived sediments settle on extensive areas of the vegetated floodplain, the conditions are quite different. Chemical and biological factors facilitate the mobilization of copper. Even a thin layer of copper-rich material may have a negative ecological impact.

4. The way in which copper affects the aquatic community is not clear. Dissolved organic carbon substances, which are present in high concentrations in the Fly River system, complex the metal and keep it in solution. Little is known about the bioavailability and toxicity of these organic copper species. Recent research has shown that these compounds may exert toxic effects comparable to the free ionic copper species.

The most sensitive biota to dissolved Cu are juvenile stages of fish and aquatic invertebrates which form the base of the trophic web. Populations of larger fish species will only show a response after their reproduction cycle has been completed. Measurable negative effects may develop with a considerable time lag.

5. The copper pollution in the Fly River floodplain is of persistent nature. Even after the cease of deposition of copperrich suspended load, the environmentally detrimental effects will continue to exist. There is no mechanism to "detoxify" the alluvial plain. Erosion processes will only remove deposits in the main river channel and on the banks immediately adjacent to it, and, to a minor degree, along the channels connecting drowned valley lakes with the main river. The mine-derived material deposited in oxbow and drowned valley lakes, and in the floodplain swamps, will remain and may be covered by less contaminated sediments carried by the Fly River in the post-mining period. However, when sedimentation rates in the floodplain return to pre-mining values, it may take centuries until copper-rich sediment deposits are sealed by an unpolluted sediment cover.

6. The possibilities to mitigate the detrimental effects of the Ok Tedi mine wastes on the Ok Tedi/Fly River system are limited when the construction of tailings and waste retention facilities is deemed uneconomical by the mining company. In order to reduce the amount of metal discharged into the environment, the recovery of copper in the mill, currently at around 85%, could be increased. This may be possible through installation of additional flotation cells, re-flotation of tailings, increase of the residence time of ore in the cells, improved control of grain size in the flotation process, sulfidization of non-sulfide ores, and other optimisation strategies. Thus, a recovery of 90% may be achievable. The cut-off grade for waste rock could be lowered which would also result in less copper reaching the fluvial environment. Measures of this kind are most probably not viable from an economic point of view; however, as an environmental protective action, they should be investigated thouroughly by the mining company.

9. ACKNOWLEDGEMENTS

The authors of the study wish to thank the United Nations Environment Programme for financial support of the research project. The study team is particularly indebted to Mr. Gerhart Schneider, Programme Officer with the Water and Lithosphere Unit at UNEP headquarters in Nairobi, who offered prompt assistance in solving all emerging problems. David Mowbray of the Department of Environmental Science of the University of Papua New Guinea (UPNG) provided invaluable support througout the various stages of the research project. The authors of the present study wish to thank him and Dr. Ian Burrows, Head of the Biology Department of UPNG, who provided working space and laboratory facilities during the field work in Papua New Guinea. Very special thanks go to Mr. Peter Gelau of Obo Station on the Middle Fly River, who helped us in accomodation and transport problems during the various sampling campaigns, and Mr. Ben Kapi Mekeo, our reliable field guide and boat pilot. We also may express our gratitude to Mr. Joseph Gabut, then with the Department of Foreign Affairs of the Papua New Guinea Government, for continued support of our research project. We wish to thank Ok Tedi Mining Ltd. for analyzing a batch of water samples and further logistical assistance.

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ANNEX

Tables A1 to A6

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2/27.3.7 Fly bank Xs. 30-32.5 cm	5300	13000	6700	5900	94400 673(0 289	147	40	19	5	کم مو		21	\$	17	27	157	4100	3 6	28	390	148	20	1.31	100
4/16.2.11 L.Dav chan./Fly 3-5 cm 4/16.2.17 19-20.5 cm	53 <i>0</i> 0 5000	13300	6900 7200	6200 6800	83600 533(92000 6440)0 461)0 760	113 124	39 32	14 21	0.6 <.4	~ 0 7 7		¥ 57	8 8	15 14	26 24	148 150	4300 4400	79 82	26 27	340 364	154 153	19 20	1.6 nd	600 bđ
4/1.11.III L. Dav.chan. 20.5-22cm	1 2500	11800	4900	5300	106200 3060	X0 185	145	36	14	0.2	0 IZ	ي. م	87	7	13	32	208	4700	8	32	335	127	20	멷	Ŗ
1/30.3.III L. Dav. chan. 26-28 cm	1 4100	12900	6200	58.00	95700 283()0 122	116	35	14	0.6	ېر اور	<u>ت</u> ۲	- 64	\$	5	26	155	4400	82	25	321	133	20	1.29	400
1/19.2.II L.Daviandu E 2.5-5 cm 1/19.2.III 5-7.5 cm	2900 2300	11600 8900	4000	5200 5200	90300 288(87500 3490	00 221 10 250	92 69	37 36	11 6	0.3	o o ⊅g		99 99 +	с н П	11 R	27 31	208 188	4200 3700	88	28	271 262	109 104	18 18	4.8 nd	600 M
1/1.11.1 L.Daviambu M 0-2.5cm	1200	3600	2300	57 00	59600 239(10 197	281	12	11	0.2	o pu		61	~	17	38	130	2300	42	15	261	87	12	- E	ġ
2/1.11.11 L.Daviambu M 2.5-5cm 2/1.11.111 14-16.5 cm	1200 100	5100 100	2400 500	4500 rd	71100 182(6500 3866	90 121 10 442	139 132	35	10 22	0.2	0 0 9 9	44	8.8	~ ~	12 16	33 20	166 38	2600 300	7 21	20	199 114	ឌា	3 14	밑멸	4
5/30.3.1 L. Daviambu N 0-3 cm > 200 µm 5/30.3.111 L. Dav. N 19.5-22 cm t	2500 900 .otal	8800 3400	4200 1900	80 00 50 00	73600 378(41100 211(00 270 00 157	183 193	35	18 12	77 77	0 0 5 7	بن سن سر سر	36	с <u>с</u>	51 61	24 21	33 23	2500	30	18 12	263 116	102 59	15 9	19.6 19.6	00 00
2/1.4.1 L. Daviambu N 0-2.5 cm 2/1.4.11 L. Daviambu N 2.5-5 cm	1200 2100	3500 7400	2100 2400	8200 6600	40100 256(49200 2340)0 276)0 288	220 226	12	19 21	.	e v g g	. r.	* 27	5 7	12 12	18 18	70 85	1200 1400	30 28	==	167 198	7 7	م ه	12 12	ra n⇔
7/24.10.111 L.Dav. S 27.5-30 cm	4300	12200	5500	3700	90400 3150	10 201	36	11	35	• •	<2+ 0	*	+ 75	2	11	37	174	5600	101	28	302	ш	18	- Ig	roi
3а/30.3.1 L. Daviambu W 0-2 сп. t 3а/30.3.1V L. Daviambu W 22-24 сп	1100	2300 10600	1200 6200	3500 5100	31500 101(90900 5890	30 86)0 1110	77 115	21 18	5 15	- 	a a g g		22 63	5 5	4 01	16 21	50 134	1000 4560	22 76	13 24	122 328	48 118	8 6T	14.5 2 Dù 1	600 đ
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3b/30.3.I L. Daviambu W 0-3 cm. ⁴ 3b/30.3.III L. Dav. W 22-24 cm t. 3b/30.3.V L. Daviambu W 39-41 cm	: 1300 700 5500	2700 1700 13000	1400 1100 6700	5500 7700 4600	30900 14 19400 16 84500 34	600 92 900 217 800 284	104 29 142	2 1 2	9 44 13	4. ^ 4. ^ 1. 4	ng ag	333	2 2 -	5 8 33 5 8 33	1007	0 T O	5.9.5		81 F0 53	11 6 5	32	9 29 1 25	6 18 6	16. 37.	2 4 2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	
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2/28.10.VI Fly R. bank 51-54 cm 20-60 µm 60-100 µm 100-200 µm	5600 9200 9600 9200	15100 12200 11200 13900	7700 5800 6000 5300	6600 8100 8800 11700	102500 47 63300 41 54600 35 52900 30	000 285 600 285 800 301 200 320	143 97 95	8 8 X 8	11 30 29	0.2 0.3 0.2	षु पु पु के	0.1	2004	16546				6 H C 6	100 12 100 12 100 12	57338 9 7 7 38 9 7 7 38	*****	3 221 3 250	3 22 6 11 3 10	<u>a a a a</u>	정정정정	
2/31.10.IV L.Pangua N 7.5-10 cm 2/31.10.V 16.5-18.5 cm	1300 3700	5200 13000	2200 6500	4000 4000	66500 21 64700 31	100 192 500 193	136 101	39 39	18 14	0.5	'면 ^杰	0.9 0.1	\$ \$	51	2 1	86.0- L.L.		87 21 97 65	200 F1	0 26 2 26	17	6 7 0 10 7	1 13	а В	열열	
3/31.10.11 L.Pangua W 2.5-4.5 cm 3/31.10.111 16.5-17.5 cm	5100 4800	18300 15800	6800 7400	5500 5200	85300 32 89100 33	900 184 800 222	116 117	88	25 2	0.9 4.0	pe pe	1.9 0.1	7	852	2 4	5 82 5	5 5	88 88 88	00 IC	8 1	33	2 15 4 14	1 17	g g	n d	
4/31.10.11 L.Pangua C 14.5-16.5c 4/31.10.111 28.5-30.5 cm	n 4 300 4200	16100 14100	7100 6400	5400 4300	109900 60 88500 61	400 324 100 334	118 135	57 59	15 31	0.2	a a	0.2	* ~	80 80	- 5	9.0	2 2	2 4	200 11 100	22 33	32	2 12 2	2 23	3.0 Nđ	520) nd	-
5/31.10. L. Pangua SE 10-12.5 cm 5/31.10. 16-18 cm 5/31.10. 23-25 cm 5/31.10. 21.25 cm 5/31.10.Y 42.5-44.5 cm	3300 3500 3400 3500	12400 12900 13300 17400 14400	6400 6600 6900 7700 7100	5900 5900 5200 5200	93900 64 96900 64 100600 67 112400 67 94300 70	200 272 400 352 800 320 300 312 300 312	111 120 148	41 45 67 33	811124	0.2 0.2 0.3	멸멸멸멸	22223	5 # 8 # 8	879830 879830	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	10,00,00,01,01 0,00,00,00,00	00000	******		32828	****		5 8 8 8 8 5 5 8 8 8 8 5	3.1 3.1	9 2 2 3 9 9	- -
1/4.4.17 L. Pangua NE 23-28 cm 1/4.4.VI L. Pangua NE 33-38 cm	3700 5400	10500 13300	5500 6900	77 00 8300	89600 42 95000 48	500 220 800 723	198	58 40	23	4. ^ 4. ^	nd nd	0.3 0.4	ლ. აჭ	23	22	2 5	14 59	39 FF	00 77	25	37.	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2 19	2,5 (C01	f 320(170(

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2/4.4.7 L. Pangua NW 32-34 cm	3900 1	15100 6	0065	6300	94500 3351	00 203	129	36	14	† ">	ų	¢.3	2# 5	~	~	50	135	430) 82	23	HC	108	19	2.21	002	
8/28.10.1 Sviss Lake 0-2.5 cm 8/28.10.11 2.5-5 cm	4500 1 5200 1	12500 7	7300	67 00	89500 3261 108900 3161	00 154 00 149	148 133	3 1	21	0.2	꼍춘	0.1 0.1	8 8 1 5		a -	2 23	201 188	520 560	88	20 26	379 417	165 185	15	22	모모	
10/28.10.II Swiss Lake 16-18.5 cm	5400	14300 6	9965	9500	119 00196	00 215	117	60	18	1.0	pq	< <	5	et.	10	53	151	450	08 0	27	350	170	20	ри	þ	
3/19.2.II Tamu Cr./Fly jct.3-5 cm 3/19.2.III 13-15 cm	5400 5400	13500 1	7100 7200	77 00 87 00	92600 463 93300 479	00 247 00 408	116	45 37	16 14	4.5 1.4	결절	33	99 33	~ ~	22	57 59	152	. 1 60	0 85 85	27 27	357 135	166 171	50 50	22	N. N	
8/30.10.1 Tamu Cr.0-2.5cm < 200µm 8/30.10 II 2.5-5 cm < 200 µm	300	1800 900	1200 900	16500 16800	27700 155 20900 144	00 355 00 260	23	8 4 EE	6 L	55	절명	2.ů	22	5 5		11	88 22	100	0 22 0	12	100 88	114	с т 90	24.7 29.4	3300 3200	
5/30.10.1 Lake Tamu II 0-4 cm 100-200 µm	2200 2600	0016	3400 2400	68.00 68.00	79700 421 50700 345	00 398 00 618	194 146	157 266	45 42	0.5	모모	0.8 1.8	5 7	~		3.6	101	330	0 22 0 26	28 24	253 208	118 104	16 12	nd Dđ	ව වි	
3/30.10.1 Lake Tamu I 0-3 cm	1300	44.00	2000	3500	52100 201	00 151	176	49	٢	0.5	ጀ	0.1	6 7	2	-	5	17	640	23	25	185	25	12	pa	ри	
6/28.10.II floodpl./Kibuz 12-14cm	5400	15000	7200	7800	96600 548	00 712	117	60	21	4 .^	nd	ç	5	0	1	5	151	. 440	0 79	27	380	174	20	1.1	1000	
1/20.2.1 Bai Lagoon NW 0-2.5 cm	1400	4600	2500	80.00	41600 173	00 146	114	%	23	0.2	р	0.2	4	5	5	16	5	1 130	0	16	142	120	6	13.5	2900	
3/20.2.II Bai Lagoon SM 1-5.5 cm	0011	2900	1600	9200	46200 171	00 140	66	56	6	0.2	рu	0.2	4	فن		1	æ) 160	16 0	14	148	108	6	ହ	臣	
1/5.4.V Bai Leg. C 23.5-27 cm	1100	6000	2900	10900	61300 257	181 00.	85	56	01	0.5	q		र उ	ې د	2	5	10	3 240	0 49	19	181	110	13	25.4	2800	
4/20.2.11 Bai Lagoon C 3-5 cm 4/20.2.111 5-7 cm 4/20.2.17 11-14 cm 4/20.2 V 14-16 cm	2100 700 1800 1300	8400 3900 9100 6600	4600 1900 4400 2900	7300 12600 6300 12800	91300 284 47500 285 100600 253 51000 313	00 224 00 243 00 243 00 295	61 28 128 121 121 121 121 121 121 121 121	43882	18 15 11	0.2 0.2 0.2 0.2	모모모모	0.2 0.2 0.5		6 7 6 5	~	*		5 36 1 30 5 1 1 2 0	0 0 33 6 48 9 3	53 53 53 53 53	261 280 280 185	138 138 158	91 61 21 13 61 21	말벌멅멅	2222	
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2/5.4.I Bai Lag. E 0-2 cm 2/5.4.II Bai Lag. E 2-4 cm 2/5.4.III Bai Lag. E 23-25 cm	3800 3100 2700	15000 13400 13000	- 6200 6900 5200	69069 0069 7100	95500 108200 111900	37500 21 33300 17 23300 12	51 12 19 19 19	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	18 23	<u>.</u>	222	233	13 44 0.4	55 55 55 55 55	** ** **	11 6	25 25 25	148 161 148	3300 4300 4100	33 88 53 33 88 53	22 23	327 349 349	122 123	20118		ط 300
5/20.2. Bai Lag. chan. 43-45.5 c	n 4900	14100	7000	12300	95200	41700 25	1 124		11	0.4	nđ	0.3	\$	57	n	12	24	150	4300	75	25	359	178	19 2	يم	800
6/20.2.II Kemea Lagoon C 2-4 cm	2600	11300	4400	7100	92300	34500 24	3 14	19 1	1 17	0.2	þ	0.2		52	-	17	25	165	3100	5	25	254	142	16 1	a q	951
3/5.4.X Remea channel 36-41 cm t.																								-	2.7 2	500
4/5.4.V Kemea Lag. W 14.5-17 cm	1200	8200	1000	3700	116500	41500 19	12	*	1 23	4 .>	pđ	0.3	₹	55	\$	12	24	175	4800	85	45	291	æ	22 0	> 6Z.	100
5/5.4.IV Kenea L. E 17.5-21 cm t.																								~	4.8 4	009
1/7.4.V Fly R. bank 33-35 cm 1/7.4.V < 2 µm	7000 8300	18500 16700	7400	7500 9100	102100 116300	(35700 21 39400 21	2 12 8 134	2 23	38	4. 4.2	모모	33	° . 4	58 74	- Q	ማ ማ	24 37	S Z	4100	82 OS	26	314	174	19 1 26 n	~ ¥ 8.д	200 T
2/7.4.III I. Kongun N 8.5-11 cm	4000	15300	6600	57 00	111200	33300 16	3 181	5	1 29	¢'>	Вđ	¢.3	\$	99	\$	10	30	111	3900	11	27	442	128	21 H	ž	-
3/7.4.V Kongun Ct. N 30-32 cm	4800	14300	7300	6500	110400	38200 18	13(0 4(1 24	0.5	Del Del	ŝ	\$	66	2	11	27	168	4800	90	28	370	155	23 J	ž	
4/7.4.III Kongun Cr. S 19-21.5 a	n 4800	14500	6600	6400	103200	33900 15	5 1L	3	16	۲. 4	\$. .	2	28	\$	80	23	143	1200	83	24	369	142	21 0	8.	100
1/23.2.1 L. Boss.SW creek 0-2.5c	n 1900	5200	1800	2400	95200	21200 9	ŋ 13	3 44	88	0.2	R	0.2	-41	80		e,	46	239	¢000	28	18	215	z	16 h	ž	
2/23.2.1 L. Boss. SW creek 0-7 C	n 2200	6700	2000	3700	94000	27200 12	8 IG	7 5() 21	0.2	'nđ	1.4	-	82	7	12	46	207	4500	5	23	221	83	16 n	ž	
4/23.2. L. Bosset island 20-60 μm 60-100 μm	1600 200 200	6100 500 800	1900 200 200	1500 400 400	107200 7000 9000	21400 4		N *** **	**5	0.2 0.2 0.2	엄덕덕	0.2 0.2 0.2	***	120 28 34		1 - 1	4°~	294 26 39	6800 2400 1000	102 53 16	7 9 7	215 40 32	នងដ	e e e 9 m v	222	
5/23.2.III L. Bosset SN 5-7.5 cm 5/23.2.IV 7.5-10 cm	2600	10300	5000	4400 5600	114800 130500	27600 14 29300 17	2 9 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	48	7 14	0.2 0.2	명명	0.2 0.2	é ri	73		14	6 8	242 274	3600	53	35	249	911	22 52	vat.) a a	

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1/24.2.11 L. Bosset NW 5-7.5 cm 1/24.2.111 7.5-10 cm	1700 1700	6000 4700	2200 2800	3500 4600	103700 28600 102000 27000	92 131	156 224	67 55	21 0	г. г. г.	ים קים	~	69 61	പ്ര പ	14 12	36 36	255 195	3200 3300	47 52	25 24	191 225	87 85	16 19 4	2 2 2	d 007
3/11.4.11 L. Bosset NW 2-4 cm 3/11.4.111 L. Bosset NW 4-6 cm 3/11.4.X L. Bosset NM 6-8 cm 3/11.4.X L. Bosset NM 18-29 cm 3/11.4.X1 L. Bosset NW 30-42 cm	2000 3900 2300	7100 5700 7200 7300	3400 2600 3900 2800	47.00 39.00 4.000 27.00	92600 31700 85600 25100 109900 27600 61000 20100	140 116 93 101	216 255 227 254	5145	32 33 33 22		៴៴៴៰		+	6 6 9 m	10	35 66 33	165 168 176 198	2600 2800 3200 3400	\$\$ \$ \$\$ \$5	18 18 25 13	244 256 291 247	85 29 29 29	117 117 119 112 112	~ 5 5 5 5 5 	40 40 40 40
2/24.2.11 L.Bosset N 2.5-5 cm	1200	3000	2000	4500	66100 18600	111	159	45	11 0	4	ب د	ت 2	42	Q	\$	32	134	2300	40	20	174	70	15 6	1	300
3/24.2.III L. Bosset N 9-11 cm 3/24.2. 10.5-12.5 cm 3/24.2.IV 24.5-26.5 cm	2009 1500 1500	6900 5900 6130	4000 3500 3200	4800 4800 5300	92000 36500 82700 27100 85900 25000	1111	269 173 156	70 44 32	10 0	4 4 4 7.7 7.7 7.7 7 7 7	ਰ ਰ ਰ ਯ ਚ ਚ	0 0 0 4 4	53 53	- 0 -	27 14	42 35 35	245 221 221	3200 2800 2900	22 23	26 21 30	264 240 251	110 101 85	18 18 17 1	d 1 2	đ 200 300
4/24.2.TV L. Bosset C 13-16 cm	2100	10000	4600	48.00	96400 30000	156	167	25	18 0).2 h	ч о	2	62	1	ព	36	227	3100	19	31	257	119	20 n	ų č	70
2/11.4.1 L. Bocset C 0-4.5 cm 2/11.4.1 < 2 µm 2/11.4.11 L. Bosset C 4.5-6.5 cm 2/11.4.11 L. Bosset C 34-36 cm	2600 2100 2200 2100	10500 9800 8800 10900	4700 4700 4400 4400	4600 5500 5300 4400	97900 32200 103800 32600 92800 25900 108400 23600	146 169 124 105	154 168 185 144	83355	5538 5538		~~~~ ~~~~~		4+ 59 65 8+ 55 8+ 52	≁ ₽ 3 3 3	01 10 7	28 11 28	173 173 181	3100 3300 2900 3860	65 60 78 78	24 21 24	285 268 313	103 94 95	20 19 22 33	4 4 4 7 2 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	40 40 40 40 40 40 40 40 40 40 40 40 40 4
4/11.4.IV L. Boss.E 23-25 cm <2µ	006E E	12900	6700	80.00	113100 73800	163	124	48	21 ~	Ē	ن. م	4	11	2	10	27	201	4100	88	27	322	117	25 n	ā T	-
1/25.2.VT Boss.chan/Fly 28-39 cc	1300	5700	2305	3500	83500 15900	- 61	128	44	15 0), 2 Di	٩ ٥.		60	1	10	54	183	3100	19	40	275	78	ų 19	ية 10	-
1/9.4.III Kai Lag. SW 15.5-18 cr 1/9.4.V Kai Lagoon SW 33-35.5 cr	1709 2100	313 0 7233	3800 3300	39 <i>0</i> 0 43 <i>0</i> 0	102000 22609 85500 32609	94	344 381	52 44	28	22 77	с. чо	89	* 58	\$ \$	8 25	35 44	214 174	3200 2600	59 59	26 23	263 225	82 79	22 B	22 777	
2/9.4.IV Fai River S 18-20.5 cm	\$300	14700	5290	6130	110200 30200	109	123	44	~ ~	*	8+ *	3	+ 62	\$	"	25	165	4200	68	28	387	138	23 1	1	900
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3/9.4.II Kai lagoon N 5-7 cm 3/9.4.III Kai Lagoon N 15-17 cm 3/9.4.III < 2 µm	3300 2700 2300	13300 13300 14800	6000 5000 6000	5800 4900 5600	112800 355 110700 336 125300 393	900 134 500 124 300 131	142 125 142	37 37	28 24	222	29 D D		333	71 67 75	3 6 6	60 00	22	534	300	2333	దారా బి.బి.బి	88.1	222 282	말드릴	알 ^ᇊ 껕	ė
1/8.4.111 Agu L. bank N 7.5-10 cz	1 4100	12200	5400	2000	112400 33	400 1 76	246	22	27	4 .>	pq	<u>ئ</u>	\$	64	\$	10	32	159	300	8	28	30	21 2	D L	þų	
2/8.4.III Agu Lake N 5-7.5 cm 2/8.4.IV Agu Lake N 7.5-10 cm 2/8.4.IV < 2 µm	4800 3600 3400	10000 11000 9800	5100 4900 5400	7200 5900 5800	122100 30 105300 31(135400 31(200 126 500 158 300 102	961 191 181	2583 2	28 26	0.5 <.4 <.4	gggg	333	হ হ হ	68 63 75	9~3	9 10	34	181	1500 1700	8 % 8	5 2 2	5553	222 222 222 222	n n n n n n n n n n n n n n n n n n n	절절절	
4/8.4.11 Agu Lake E 5.5-8 cm 4/8.4.111 Agu Lake E 15-16.5 cm	3800 4400	13000 14000	600 6500	5100 6900	115700 26. 117600 33	900 125 900 125	164 1146	57 49	26 24	10 10	ष्ट्र	î.	22	64 65	~ ~	æ 01	33 29	173	200 100	88	30 ¥	1 8	2 2 2 2		N Pd C2	0
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3/16.2.1 Fly R. bed/0bo 0-4.5 cm 3/16.2.111 25.5-27.5 cm	10000	34400	9700 6	51600 { 51000 {	32800 4 18200 3	12300 80 18500 80	13 237 12 270	626	93 117	0.9 0.9	150+ nd	1.0	204	41 52	58 13	6 14	15	179 2 114 2	600 3	513	38	9 4 64	1 12	nd 0.6	nd 3500	
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5/27.3.111 Fly bank XS. 20-22 cm 20-60 μm 5/27.3.V Fly bank XS. 30-32 cm 5/27.3.VI Fly bank XS. 46-48 cm 5/27.3.VI < 2 μm	13980 9500 9120 9120 9120 9120 9120	31100 40300 35200 36100 20100	8600 6500 9000 8700 10600	48700 38000 47800 38500	80960 73160 83300 81160 101000	45900 1 24400 1 42100 6 41000 6 57900 7	3 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	1 2 2 7 7 2 8 2 7 7 2 8 2 7 7 2 8 2 7 7 2 8 2 7 7 2 8 2 8	77 17 28 91 10 11	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	*	20 10 8 4 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5	12224	\$23859 \$	54 ° ° ° 1	11 17 16 23	8 2 2 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 2 1 3 1 3	2900 1600 2600 3300	42 38 33 342 45 47 38 38 37 45	32 53 13	425 529 441 416	凝킄킄탿뗧		1	4 0 0 0 0 4 0 0 0	
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4/16.2.1 L.Bav. chan./Fly 0-2 cm 20-60 µm	9600 12000	31000 27800	7300 6100	25300 25400	83400 73300	44100	742 1 520 1	30 S	22 83	55 0. 50 0.	2 9 2			6 7 11	16 11	15	18 15	11 115	3200 2400	57	29 19	415 487	335 387	1	с. Г	q 500	
1/19.2.1 L.Daviandu Ost 0-2.5cm 20-60 µm	4600 6800	17300 12600	4700 2700	91 00 78 00	87700 47800	31000	(33 1 (31 1	[<i>1</i> 66	ខេន	50 50	с, с ,	12 12		8 2	ς, φ	12	27 15	90 198	3600 1900	11	34	327 256	156 149	18	2. P	1 00	
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4/1.4.11 L. Daviandu E 6-8 cm 4/1.4.111 L. Daviandu E 12-14 cm 20-60 µm 4/1.4.V L. Daviandu E 20-22 cm	9800 8600 13400 7700	36500 30300 37200 32100	7200 8300 7100 7900	37800 52900 41000 48200	80400 78900 75900 70400	37600 42500 255900 35300	526 526 555 533 2 533 2		8 2 2 2 2	0000 8 # # 8	2 2 2 2 8 2 2 2 2 2		2322 2322	33 47 47	24 29 23	80 J ~ 4	16 12 12	118 121 125	2500 2800 1800 2100	23 41 33	27 25 24	455 430 530 394	385 388 388 388	2261	ы 1.65 5.94 1.94 33	90 90 90	
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2/19.2.1 L.Pangua SW 0-2.5 cm	4300	17700	6100	7600	90400 4	5200 48	0 18	246	5	0.3	몓	1.0	10	65	16	18	Ĩ	221	3500	68	35	318	186	17	ער ער	Ţ	
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4/19.2.1 Oriv bed N'L.Kibuz 0-3 20-60 µm 4/19.2.11 3-6 cm 4/19.2.1V 22-24 cm 4/19.2. 26-28 cm 4/19.2. 33-35 cm 4/19.2. 33-40 cm 4/19.2. 43-45 cm	 25 8800 12100 12100 9100 9100 6800 6450 9100 8300 	36500 37100 42200 44600 44600 44600 29600 26600 26600 26500 26500 26500	8800 5300 6300 8400 6800 6100 7600 7100 7200	53600 46500 61700 19400 12500 11300 11300 11300 11000 10000 8200	78200 66300 86900 88900 79400 90400 92500 79800 92500 97800	45200 40360 51060 54200 54200 54200 57500 45000 45000 45000	539 517 517 517 517 517 517 517 517 517 517	294 8 247 4 247 4 279 8 279 8 1999 9 1499 6 1499 6 1499 6 1493 2 201 4 4 201 4 4 155 4 155 4	1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	10380000880 252555555555	+00		21 33 42 42 42 42 42 42 42 42 42 42 42 42 42	88135685568555 8913668568555	13 13 13 13 13 13 13 13 13 13 13 13 13 1	22232255 222255 22255 2255 2255 2255 22	151 151 162 163 164 165 165 165 165 165 165 165 165 165 165	2400 1600 2100 2500 2500 1800 3300 3300 3200 3200 3100 3100	9		396 414 447 506 538 494 429 429	477 465 356 356 356 356 356 356 356 356 356 3	***	ла ла ла ла ла ла ла ла ла ла ла ла ла л	패션 3300 2900 개년 00 개년 10	
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3/5.4.IV Kence channel 18-20 cm 3/5.4.V Kence channel 25-27 cm	9600 6500	52300	7400 8000	41300	79400 25 93400 29	700 404 000 257	165 167	524 432	72 108	<.4 0.7	nd Dd	ن. م م	1 28	33	5	10 16	96 36	1500 2500	26 39	22 28	526 465	301 301	8 2	4.51 J	00
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4/9.4.I Kai Lagoon M 0-2.5 cm 4/9.4.II Kai Lagoon M 2.5-5 cm	7500 6600	35100 29100	6900 6300	5800 6200	93700 4 92600 4	4400 2. 0500 15	87 F1	69 4. 52 2 ⁷	78 22 7	v v v v	4 17 4 19	0+ 0. 0+ 0.	4 9 9	+ 47 + 49	20 24	r~ 00	18 20	142 160	260 310	0 44 0 48	26 26	468 843	229 223	51 M.	2.22 1.96	800 100	
2/8.4.1 Agu lake N 0-2 cm > 200 µm 2/8.4.11 Agu lake N 2-5 cm	3500 2400 3900	13000 7700 11600	5300 3400 5200	5800 4900 5600	95200 2 58700 2 107800 4	28200 1 26100 21 28800 21	1 1 5 28 4 0 28 4 0	2335	681	~ ~ ~ ~	4 4 4 9 9 9	.		51 63	5 5 11	8 8 11	3 2 12 25	144 93 160	290 390	0 59 0 39 76	28 18 29	318 206 322	146 88 127	17 12 19	말말말	222	
4/8.4.I Agu Lake E 0-2.5 cm 4/8.4.I < 2 μπ	7900 4200	36000 18700	7300 8000	4500 5500	99700 2 115600 3	38200 1 19900 2	44 1	76 (6 6. 2	5 10		6 4 6 16	46 46	12	r 6	20	153	260 350	0 39	28 33	474 384	237 201	51 61	1.94 nd	200 Bđ	
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6/8.4.IV Agu R. S. охрож 20-23с1 20-60 µm	m 11200 14300	51400 46700) 7600 1 6000	58000 53400	81900 1 70300 2	35900 5 21600 3	53 2 10 1	24 3	52	2 5	2 Z			31	34	യന	15	105 78	130	0 45	24 13	530	473 471	5 4	nd nd	nđ	
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W1/22.10. Fly R,/Obo W11/22.10. Fly R./Obo W1/24.10. Fly R./Obo W2/28.10. Fly R./Nbuz W2/16.2. Fly R./Nbo W1/21.2. Fly R./Obo W1/21.2. Fly R./Obo W1/10.4. Fly R./Obo W11/10.4. Fly R./Obo	26.1 26.1 26.4 27.3 29.6 29.6 29.2 29.2 29.2 29.2 29.2 29.2	7.7.8.0 7.7.9 7.7.8 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.9 7.7.7 7.7.9 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7 7.7.7	148 145 137 172 130 130 132 132 132	60 23 54 4 55 3 3 6 6 6 4 6 5 6 7 3 3 4 6 6 0 4 6 5	22222222222222222222222222222222222222	1570 2270 1420 1320 1320 1420 1420 1420	ла ла 565 722 722 722 722 722 722 722 722 722 72	41200 34300 26000 39900 222100 222100 2225900 nd 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1450 11520 1190 11040 11040 11290 11290 11290 11290	a g 128 2 3 3 5 1 1 1 2 3 3 5 3 3 5 1 3 1 2 3 3 5 1 3 1 2 3 3 5 1 1 2 3 1 3 1 2 3 1 3 1 2 3 1 3 1 2 3 1 3 1	6670808484 889418188855	<u> </u>		81100022256199 811010522556999		ម៉ំសីដ¥ស្សំស្អុ៤គ 。	ри и и и и и и и и и и и и и и и и и и	55555555555555555555555555555555555555
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rabuuli briuge W1/14.4. Ck Tedi, W2/14.4. Tabubil bridge W3/29.2. Ok Tedi/Ok Menga	nd nd	7.9 7.9	300 2 4 2 204 1	96 16 p	5889 4513 10610	4 430 4370 3660	1790 1780 nd	55700 44700 35500	2550 2430 2140	388 46 337 63 316 10	80 27 33	r & &	<u></u>	32 21 6	444	17 17 26	158 124 120	16.0 16.6 9.4
W1/29.2. Ok Menga W2/29.2. creek/Ok Menga	pu pu	7.1	186 1 197 1	g g	146 2	1080 2720	nd	38400	2330 1210	418 96 146 54	11	19 8	<.1 <.1	8 8	rd d	ъъ	nd rd	1.5 1.6
Table A4. Analyses where appropriate.	ц Ц О	TY F	liver	and	ð	redi	wat	ers.	Med	lans,	mean	s and	sta	ndarı	ц Се	riat i	ទី	only given

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W3/30.3. Lake Daviambu W	34.5 5.7	43]	145 nc	1 19	60 <2	9 0	420 54	-0 1	2 2	50	~ 68	ۍ م		У 2	\sim	₹ 	2	0	ୁ ଜୁନ	002 n	0	1 61
W2/4.4. Tamu Creek mouth/Fly	29.2 6.0	45	5 DK	1 15	60 14	1 91	190 6;	72 4	22	50	82	5		4	2	~	~	0 1	40 0	.004 0	01 0	92 11
W1/5.4. Bai Lagoon C	29.4 6.2	27	65 I:	3 10	20 15	36 5	890 3;	71 3	4	77 2	72 <	ា ទ	~ ~	Ľ	~	⊽ 1	-	0	60 0	0250	005 0	н П П
W3/5.4. Kemea Lagoon W	29.6 5.6	18	40 3(1 12	20	2	010 4	96]	ω.	72 1	95 3	=	.0	34	~	۲ ۱		6 0	30 0	.012 0	.015 0	15 L
W4/5.4. Kemea Lagoon E	30.6 5.4	14	11 NI	10	20 6	1	080 4.	18	3	65 19	40 3	2	~	~	~	√ √		с. О	50 0	.003 n	0	23 11
W2/7.4. Lake Kongun N	31.2 6.4	22]	102 NI		41 15	11 3	350 34	49 2	-	02 20	00 2	₹ 1	~	v L	~	√ 1		5 0	10 n	d D	ŭ	_
W3/7.4. Lake Kongun S	28.6 5.8	26	3 M	9	93 19	<u> 9</u> 3	960 3.	76 2	4	83 21	00		 	~	~			о 6	15 0	a 010.	ŭ	_
W4/7.4. Lake Kongun	28.2 5.8	28	5 14	1	15 2(16 4	240 3;	72 2	× 10	50	72 <	ۍ. 	~	- -	~	₹		40.	15 0	.003 0	02 0	38
W2/8.4. Agu Lake	31.4 6.5	44	67 nc	1 11	50 35	32 7	870 4	7 66	1	62 1	37 <	⊽ ശ	\$ \$	74	~	⊽ -	5	4	14 n	ц ц	ž	
W1/9.4. Kai Lagoon SW	29.2 6.4	30	65 1(11	30 25	54 4	380 4	76 2	*	50 2	41 <	5	v		Ų Ų	₹ L		و. 0	02 n	n q	0	۔۔ چ
W11/9.4. dto.			N	11	20 28	24	350 4	36 2	ς Υ	50 12	10	~, ~		-	~	√ 1						~
W1/11.4. Lake Bosset SW	29.3 6.0	21	57 NI	11	70 30	37 3	380 4	84	Ω.	65 22	70 1	₹ 1	~	_	~	~		0 6	15 0	.012 0	.015 0	4
W3/11.4. Lake Bosset NW	28.7 6.4	38	50 19	11 6	20 25	14 6	200 5;	21	× و	50 1	24 <	ي ج	~	` _	~	1	~	1 0.	20 0	.0250	.015 0.	67
W2/20.2. Bai Lagoon W	30.5 6.5	21	98 33	9 12	80 nc	•••	060 21	84	ŝ	ud Dd	85	1	~		~	₹ 1		Д О	-	ц П	0	2 1(
W1/23.2. L. Bosset Creek	31.1 5.9	32	37 2.	3	140 nc	с Г	980 8.	75 (ģ	nd 14	40 ~	5 1(, ,		~	~ _	-	1 I	и 	u D	ž	Ĕ
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BLANGARG GEVIATION ONLY GIVEN alla a U U D H ñ where appropriate. NF means unfiltered water sample.

sample	temp. pH	БО	id. oxy.	SS	Na	X	ca Mg	Sr	Al	Fe F	ln 2n	3	ខ	đ	Ŵ	HCO ₃	NH4	N_2	HS	S,	ğ	
	ပ	\Sil	* 5	/bu	1 µg/]											1/ <u>5</u> 1						
W2/15.2. Kinda Ch. flow -> river	28.7 6.	8 46	47	Ξ.	1 1340	pu (7520 486	47	pu	302	23 34	ć. 1	8	\Box	ŝ	20	pu	pu	ŋğ	0.8	pu	
M2/3.4. JASELEI (FEEK	29.7 6.	4 95		g	1840	144	17300 1130		<20 <	335	38 15	<.1	2	4	Ð	3	0.5	0.01	50.00	51.8	6	
W3/15.2. LOWER FIY/OGWA	28.2 7.	8 151	75	វ័	8 2280	pu (26400 2210	200	pq	44	<5 <5	<u>.</u> 1	Q	4	φ	Ъ	P	멑	þ	рг	рц	
W1/15.2. Lower Fly/Ogwa	28.6 7.	6 153	74	5	4 2320	pu	26700 2120	<u>1</u> 98	рq	37	8	<u>.</u> .	-	4	ĉ	pd	рq	Ы	pu	pu	pu	
W3/3.4. Lower Fly/Ogwa	29.9 7.	7 153	65	pu	2350	1 405	27600 2120	175	<50	20	11 <5	<u>.</u> .	ŝ	4	ŝ	ន	<.05	g	pq	7.3	-	
W1/3.4. Strickland River	29.2 7.	7 162	74	18_{-}	4 3360	1 426	27400 2920	187	119	43	<5 14	ć.1	Ç	4	ŝ	8	b	р	р	9.9	ف	
W1/16.2. Strickland River	27.5 7.	6 156	11	116	4 2840	pu	26200 2670	213	pq	242	8		2	4	(5	84	Ъд	Ъ	P	8.8	2	
W1/27.3. floodblain	13 B 6	5 106	30	1	1460	DAR.	10000 007	112	, Eû	707	16 J2	-	c	2	ų	ξ	((0	C C			
pond Obo		2	5	1	б. Г	5		24	Ś	500	C, C,		0	7	9	3	n	0.0	To . o G	10.0	ת	
W2/28.3. floodplain	28.0 6.	4 280	29	46	2480	811	50000 2840	27.7	<50	3845	971 18	0.0	07 07	û	10	162	0.6	0.00	8 0.02	3.0	11	
w1/26.10. floodplain	29.0 6.	5 661	29	ри	2440	pu (135700 536	0826	pq	<10 2	290 21	0.3	14	Ļ	65	рц	pu	рq	рц	pq	pq	
seepage/Obo W2/26 10 flondplain	y Pu	1 264	14	jų ir ir	120	Г. И	52200 1060	107	ť.	216	01 313	ر د	7	2	, t	7	7	-				
seepage/N'Obo	5		1	2		21		10.7	nii	010	6T 0/0	7.0	7	7		DI	2	2	2	2	IJ	
W2/1.4. floodplain seenare/W/Obo	27.8 6.	2 166	11	рц	1590	1 371	30600 1870	168	<50	810	249 19	pu	9	ри	6	91	0.25	0.00	30.15	1.9	11	
W1/28.10. pond/F1Y N'Obo	28.8 6.	5 132	38	18	1490	pq -	25600 1180	153	nđ	145	5 24	<.1	22	\$	12	рц	pu	Ы	ри	pu	рц	
W1/29.10. pond/	nd 5.	g 67	42	pu	1720	ри	13200 341	82	pq	48	49 34	<.1	4	Ω	ŝ	рq	рг	P	рц	рц	nd	
rloodplain Ubo W2/29.10. floodplain	nd 7	J 158	pu	ри	1950	pu	31400 920	191	pu	<10	<5 <5	<.1	ŝ	Ą	01>	þu	pu	pu	pr	pu	pu	
drain Obo W1/30.10. floodplain	nd 6.	l 151	43	ЪП	1610	p	30700 1760	209	pu	600	247 24	0.3	15	2	<10 <	pu	pu	p	je L	6.0	E P	
seepage N'Obo													1	1				ļ	•	}	1	
W1/30.3. Lake Daviambu channel	28.9 6.	5 106	5	ۍ ي	1350	350	20500 1040	113	<50	114	122 8	0.0	ۍ ۲	1	\$	65	0.15	0.01	5 0. 02	0.57	ß	
W1/1.4. LaKe Daviambu channel W2/22 10 I Daviambu channel	28.6 6. 76.7 7.4	1 95 1 95	4 (pu v	1340	- 289	18200 950	85	144	497	38 26		4.	Δ.	ა:	22	0.15	0.01	0.02	Ъ,	10 10	
W2/24.10. L. Dav. channel	26.5 7.8	3 137	3 3	212 212	1390	22	26000 1210	162	pi pi	t Ê	9 ℃ 9 ℃	50	3~	90	= 5	ם קיים		n d		D D D	קר קר	
W1/19.2. Lake Daviambu NE	29.4 7	3 140	70	55	1640	рц	26300 1260	171	l D	42	4 - 2 - 3	: 7	14	r pu	i o	l P	a p	n i	a P	6.2	엄	
W4/30.3. Lake Daviambu NE	32.2 6.	3 81	95	pu	1420	234	14800 885	83	<50	195	28 <5	1	$^{\circ}$	û	ŝ	47	0.15	0,00	5 nd	nd	pu	

(cont.)

sample	temp. pH	con	l. oxy.	SS	Na K	Ca	łg Si	R .	e.	문	2u Z	ਤ	3	ୟ	9	S N N	H4 N	୍ୟ ଟ	S S	×	
	c.) রা	*	T/pm	µg/1											<u>g/1</u>					
W3/1.4. Lake Daviambu NE	30.0 5.	80	17	pu	1330 241	14100	838	ور ع	111 0	23	19	<.0 4	\$	û	ĉ	46 0	.40	.012 0.	01 0.4	11 1	
W4/22.10. L. Daviambu E	nd 8	n ng	8	114	1740 nd	32300	1390 10	97 nd	(10 (10	ŝ	11	2	14	J	15 r	ч р	ч г	р р	l nd	пđ	
W5/22.10. L. Daviambu C	29.0 8.	9 155	144	2	1910 nd	30700	1380 1	92 68	<u>1</u>	\$5	13	<u>.</u>	ŝ	₽	(10	d n	ч р	d D	l nd	ŋđ	
W5/30.3. Lake Daviambu C	nd 7.	1 78	11	Ы	1300 <20	14800	823	t4 €2	0 141	ŝ	19	5	ഹ	4	ΰ	45 0	.15 0	.000.0	01 nd	თ	
W4/1.4. Lake Daviambu SE	32.1 8.	2	140	nđ	1340 124	12200	633	51	11 0	\$ 5	ŝ	<u>.</u> 1	4	ç	ۍ	38 0	.15 0	.002.0.	01 nd	80	
W3/24.10. L. Daviambu SE	30.1 8.	153	127	ő	1450 nd	30400	1250 1	53 Dd	<10 <10	ĉ	16	ć.1	ŋ	ц	(10 1	d n	u p	d B	рш П	pu	
W4/24.10. L. Daviambu SE	32.8 7.	170	115	ഹ	2700 nđ	29900	1580 21	pu oc	185	ۍ	14	` ''	ഹ	ŋd	17 T	d n	d D	d nc	P	Ъĩ	
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Table A6. Analytical results for mixed waters. Medians, means and standard deviation calculated only for filtered waters of the Middle Fly region.

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